

Physics FSC Part 2 Online MCQ's Test

Sr	Questions	Answers Choice
1	Laser is a beam of light which is	A. Monochromatic B. Coherent C. Unidirectional D. All of these
2	Helium Neon Laser Beam emitted from discharge tube has a colour.	A. Blue B. Green C. Red D. Black
3	Laser can be made by creating.	A. Meta stable B. Population inversion C. Excited state D. All of these
4	When meta l is heated sufficiently electrons are given off by the metal. This phenomenon is known as.	A. Photoelectric effect B. Piezo electric effect C. Thermionic emission D. Secondary emission
5	Bremsstrahlung radiation are examples of	A. Atomic spectra B. Molecular spectra C. Continuous spectra D. Discrete spectra
6	X- ray diffraction reveals that these are	A. Particle type B. Wave type C. Both wave and particle D. None of above
7	Kx -Xrays are produced due to transition of electron from.	A. K to L shell B. L to K shell C. M to K shell D. M to L shell
8	In Helium Neon laser, discharge tube is filled with Neon gas.	A. 10% B. 15% C. 85% D. 90%
9	The rest mass x ray photon is	A. Infinite B. Zero C. 1.67×10^{-17} kg D. All of the above
10	Which is not true for X rays	A. X rays are not deflected by electric field B. X rays are polarized C. X rays consist of electromagnetic waves D. X rays can be diffracted by grating
11	Production of x rays is reverse process of	A. Photo electric effect B. Compton effect C. Anihilation D. Pair production
12	Photons emitted in inner shell transition are.	A. Continuous X-rays B. Discontinuous X rays C. Characteristic X rays D. Energetic X rays
13	In an electronic transition atom cannot emit.	A. Infrared radiations B. Visible radiations C. Ultraviolet radiations D. Gama radiations
14	An electron in H -atom is excited from ground state $n=4$, How many spectral lines are possible in this case.	A. 6 B. 5 C. 4 D. 3
15	Radius of first Bohr's orbit is.	A. 0.053 nm B. 0.053 mm C. 0.053 micro meter D. 0.053 cm

D. 0.053 m

16 The following gas was identified in the sun using spectroscopy

- A. Hydrogen
- B. Helium
- C. Carbon
- D. Nitrogen

17 We can find from de Broglie formula

- A. Wavelength
- B. Amplitude
- C. Speed of wave
- D. Frequency of wave