

NAT I Medical Chemistry

A Lee bath B. Water bath C. Wire gauze D. Thermostat 2 Which one is not usually used for the crystallization 3 Which one is the property of an ideal solvent 4 Should be expensive B. It should react chemically with the solute Solute C. Impurities should crystallize along with the solute D. Should be safe to use 4 Gooch crucible used to filter the solution of 5 Gooch crucible is used for the filtration of precipitates having 6 By using the fluted filter paper rate of filtration is 7 During the folding of filter paper the apex form angle of about 8 The pore size of the filter paper depends upon A Nature of the medium D. Mass of the particles D. Mass of the particles D. Mass of the particles	Sr	Questions	Answers Choice
8. Acetic acid C. Sulphuric acid D. Chloroform A. Should be expensive B. It should react chemically with the solute C. Impurities should crystallize along with the solute D. Should be safe to use A. H≺sub>2SO ₄ B. HCI C. KMnO ₄ B. HCI C. KMnO ₄ B. HCI D. Both B & amp; C A. High solubility B. High concentration C. High lignition temperature B. Horeased B. Decreased C. Filtration is constant D. Having no effect A. 80 ^o B. 60 ^o B. 60 ^o C. 180 ^o D. Both B & B. 60 ^o D. Horeased B. Decreased C. Filtration is constant D. Having no effect A. 80 ^o C. 180 ^o D. 90 ^o D. 90 ^o D. 90 ^o A. Nature of the medium B. Temperature of the medium C. Size of the filter particles	1	If the solvent is inflammable for heating purpose we use	B. Water bath C. Wire gauze
B. It should react chemically with the solute C. Impurities should crystallize along with the solute D. Should be safe to use 4 Gooch crucible used to filter the solution of B. HCI C. KMnO A. H _{2 H_{2 KMnO Sub>4 Sub>4}}	2	Which one is not usually used for the crystallization	B. Acetic acid C. Sulphuric acid
4 Gooch crucible used to filter the solution of B. HCI C. KMnO ₄ D. Both B & Amp; C A. High solubility B. High concentration C. High ignition temperature D. Low temperature D. Low temperature D. Low temperature D. Having no effect A. 80 ^o D. Having no effect A. 80 ^o D. Having no effect A. 80 ^o D. 90 ^{o<td>3</td><td>Which one is the property of an ideal solvent</td><td>B. It should react chemically with the solute C. Impurities should crystallize along with the solute</td>}	3	Which one is the property of an ideal solvent	B. It should react chemically with the solute C. Impurities should crystallize along with the solute
5 Gooch crucible is used for the filtration of precipitates having B. High concentration C. High ignition temperature D. Low temperature A. Increased B. Decreased C. Filtration is constant D. Having no effect 7 During the folding of filter paper the apex form angle of about 8 The pore size of the filter paper depends upon B. High concentration C. High ignition temperature D. Low temperature A. Increased B. Decreased C. Filtration is constant D. Having no effect A. 80 ^o C. 180 ^o D. 90 ^o D. 90 ^o A. Nature of the medium B. Temperature of the medium C. Size of the particles	4	Gooch crucible used to filter the solution of	B. HCI C. KMnO ₄
B. Decreased C. Filtration is constant D. Having no effect A. 80 ^o B. 60 ^o C. 180 ^o D. 180 ^o D. 90 ^o A. Nature of the medium B. Temperature of the medium C. Size of the particles	5	Gooch crucible is used for the filtration of precipitates having	B. High concentration C. High ignition temperature
7 During the folding of filter paper the apex form angle of about 8 E. 60 ^o C. 180 ^o D. 90 ^o A. Nature of the medium B. Temperature of the medium C. Size of the particles	6	By using the fluted filter paper rate of filtration is	B. Decreased C. Filtration is constant
8 The pore size of the filter paper depends upon B. Temperature of the medium C. Size of the particles	7	During the folding of filter paper the apex form angle of about	B. 60 ^o C. 180 ^o
	8	The pore size of the filter paper depends upon	B. Temperature of the medium C. Size of the particles