

NAT I Medical Chemistry

2 Which of the following does not reflect the periodicity of elements? B. Electronegativit C. lonisation poter		Answers Choice	Questions	Sr
Which of the following does not reflect the periodicity of elements? B. Electronegativit C. Ionisation poter D. Neutron/proton A. F. B. Fr. C. S. D. Be Which of the following has greatest tendency to lose electron? A. Oxygen B. Chlorine C. Nitrogen D. Fluorine The alkali metal which is liquid at 15°C is A. K. B. Cs. C. Na. D. None Which of the following pairs are chemically dissimilar? A. Na and K. B. Ba and Sr. C. Zr and Hf.		B. Ammonia C. alcohol D. Kerosene	Potassium is kept in	1
Which of the following has greatest tendency to lose electron? B. Fr C. S D. Be A. Oxygen B. Chlorine C. Nitrogen D. Fluorine The alkali metal which is liquid at 15°C is A. K B. Cs C. Na D. None A. Na and K B. Ba and Sr C. Zr and Hf	vity ential	A. Bonding behaviour B. Electronegativity C. Ionisation potential D. Neutron/proton ratio	Which of the following does not reflect the periodicity of elements?	2
4 Which of the following elements is most electronegative? 5 The alkali metal which is liquid at 15°C is 6 Which of the following pairs are chemically dissimilar? B. Chlorine C. Nitrogen D. Fluorine A. K B. Cs C. Na D. None A. Na and K B. Ba and Sr C. Zr and Hf		B. Fr C. S	Which of the following has greatest tendency to lose electron?	3
5 The alkali metal which is liquid at 15°C is B. Cs C. Na D. None A. Na and K B. Ba and Sr C. Zr and Hf		B. Chlorine C. Nitrogen	Which of the following elements is most electronegative?	4
6 Which of the following pairs are chemically dissimilar? B. Ba and Sr C. Zr and Hf		B. Cs C. Na	The alkali metal which is liquid at 15°C is	5
		B. Ba and Sr C. Zr and Hf	Which of the following pairs are chemically dissimilar?	6
7 Veriable valency is generally exhibited by	ements ents	A. Normal elements B. Transition elements C. Metallic elements D. None of these	Variable valency is generally exhibited by	7
8 Which has largest first ionization energy? B. Na C. K D. Rb		B. Na C. K	Which has largest first ionization energy?	8