

MDCAT Chemistry Chapter 6 Chemical Equilibrium Online Test

Sr	Questions	Answers Choice
1	When the concentration of product is increased the instantaneous rate of reaction with reference to reactants will be	A. Positive B. Negative C. the same D. falling curve
2	The concentration of product is increasing from 30 mole/dm ³ to 40mol/dm ³ in 0.5 sec then rate of reaction will be-----mole dm ⁻³ sec ⁻¹	A. 0 B. 20 C. 15 D. 25
3	Consider gas is measure in bars then the units of rate of reaction is	A. Mole dm ⁻³ sec B. Bars sec C. Mole dm ⁻³ sec ⁻¹ D. Bars sec ⁻¹
4	The reaction kinetics concerned with the	A. Rate of reaction B. Direction of reaction C. Factor effecting rate of reaction D. both a & b
5	Reaction kinetics is important to discover the---under which reaction will proceed most economically:	A. rate constant B. Conditions C. volume D. equilibrium point
6	Rusting of iron is the example of	A. Fast B. Slow C. moderate D. depends upon conditions