

Everyday Science Online Test

| Sr | Questions | Answers Choice |
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| 1 | Which elements are called bridge elements? | A. Element of the third period B. Element of the second period C. Element of the first group D. Element of the second group |
| 2 | Which of these is not a noble gas? | A. Argon B. Hydrogen C. Krypton D. Neon |
| 3 | Who disproved the 'plum pudding' model of the atom? | A. Dalton B. Cavendish C. Ernest Rutherford D. Bohr |
| 4 | What is the only element that normally has no neutrons? | A. Oxygen B. Hydrogen C. Krypton D. Neon |
| 5 | Who disproved the Rutherford model of the atom? | A. Dalton B. Cavendish C. Langumir D. Bohr |
| 6 | What is an empirical formula? | A. the mean mass of the atoms of an element B. The simplest ratio of atoms of each element in a compound C. the sum of protons and neutrons in the nucleus D. The actual numbers of atoms of each element in a compound |
| 7 | German silver is an alloy of | A. Copper: 55 nickel: and zinc : 20 B. copper: 25 nickel: 45 and zinc : 30 C. copper: 25 nickel: 55 and zinc : 20 D. copper: 15, nickel: 45 and zinc: 40 |
| 8 | What is the percentage of silver in German Silver (also known as nickel Silver) | A. 100 B. 0 C. 50 D. 25 |
| 9 | Which of these metals has the highest density? | A. Platinum B. Lithium C. Tungsten D. Iridium |
| 10 | Which one of the following metals does not react with water to produce hydrogen: | A. Sodium B. Potassium C. Lithium D. Cadmium |