

Chemistry Fsc Part 2 Chapter 9 Online Test

Sr	Questions	Answers Choice
1	Cyclic structure of benzene was proposed by	A. Dewar B. Faraday C. Down D. Kekule
2	Effect of substituent on benzene ring is due to	A. Resonance B. Inductive effect C. Both a and b D. Neither a nor b
3	Biphenyl is	A. Monocyclic aromatic compound B. Polycyclic aromatic compound C. Polycyclic fused ring compound D. Alicyclic compound
4	The number of possible isomers of xylene are	A. 2 B. 3 C. 4 D. 5
5	The conversion of n-hexane into benzene by heating in the presence of Pt is called	A. Isomerization B. Aromatization C. Dealkylation D. Rearrangement
6	Aromatic compounds burn with sooty flame because.	A. They have high percentage of hydrogen B. They have a ring structure C. They have high percentage of carbon D. They resist reaction with air
7	Which compound is the more reactive	A. Benzene B. Ethene C. Ethane D. Ethyne
8	Among the following the compound that can be most readily sulfonated is.	A. Toluene B. Benzene C. Nitrobenzene D. Chlorobenzene
9	Benzene can not undergo.	A. Substitution reactions B. Addition reactions C. Oxidation reactions D. Elimination reactions
10	Aromatic hydrocarbons are the derivatives of	A. Normal series of paraffins B. Alkene C. Benzene D. Cyclohexane
11	The benzene molecule contains.	A. Three double bonds B. Two double bonds C. One double bond D. Delocalized sigma electron charge
12	Which compound is the most reactive one	A. benzene B. ethene C. ethane D. ethyne
13	During nitration of benzene, the active nitrating agent is	A. NO_3^- B. NO_2^+ C. NO_2^- D. HNO_3
14	Amongst the following, the compound that can be most readily sulfonated is	A. toluene B. benzene C. nitrobenzene D. chlorobenzene
15	Which one is not a meta directing group	A. $-\text{COOH}$ B. $-\text{CHO}$ C. $-\text{COR}$ D. $-\text{SO}_2\text{R}$

D. -NH₂

16 Amongst the following, the compound of that can be most readily sulphonated is

- A. Toluene
- B. Benzene
- C. Nitro-benzene
- D. Chloro-benzene

17 m-choronitro benzene is prepared by

- A. Nitration of chlorobenzene
- B. Nitration of Benzene
- C. Chlorination of Nitrobenzene
- D. Nitration of m-chlorobenzene