

Chemistry Fsc Part 2 Chapter 8 Online Test

Sr	Questions	Answers Choice
1	Synthetic rubber is made by polymerization of.	A. Chloroform B. Acetylene C. Divinylacetylene D. Chloroprene
2	Which one of the following gases is used for artificial ripening of fruits.	A. Ethene B. Ethyne C. Methane D. Propane
3	The addition of unsymmetrical reagent to an unsymmetrical alkene is in accordance with the rule	A. Hund's rule B. Markownikov's rule C. Pauli's Exclusion Principle D. Aufbau Principle
4	Vinyl acetylene combines with HCl in	A. Polyacetylene B. Benzene C. Chloroprene D. divinyl acetylene
5	Preparation of vegetable ghee involves.	A. Halogenation B. Hydrogenation C. Hydroxylation D. Dehydrogenation
6	The presence of a double bond in a compound is the sign of.	A. Saturation B. Unsaturation C. Substitution D. None of these
7	Which one of the following gases is used for artificial ripening of fruits	A. Ethene B. Ethyne C. Methane D. Propane
8	When methane reacts with Cl_2 in the presence of diffused light the products obtained are	A. Chloroform only B. Carbon tetrachloride only C. Chloromethane and dichloromethane D. Mixture of a, b, c
9	The addition of unsymmetrical reagent to an unsymmetrical alkene is in accordance with the rule	A. Hund's rule B. Markownikov's rule C. Pauli's Exclusion Principle D. Aufbau Principle
10	Vinyl acetylene combines with HCl to form	A. Polyacetylene B. Benzene C. Chloroprene D. Divinyl acetylene
11	Formula of chloroform is	A. CH_3Cl B. CCl_4 C. CH_2Cl_2 D. CHCl_3
12	Which gas is used for artificial ripening of fruits	A. Ethene B. Methane C. Propane D. Ethyne
13	Synthetic rubber is made by polymerization of	A. Vinylacetate B. Acetylene C. Divinylacetylene D. Chloroprene
14	Which compound is the most reactive	A. Benzene B. Ethene C. Ethane D. Ethyne
15	Vinyl acetylene react with HCl to form	A. Polycetylene B. Benzene C. Chloroprene D. Divinyl acetylene

D. Divinylacetylene

16 Which one is not property or uses of mustard gas

- A. Used in 1st world war
- B. Powerful vesicant
- C. High boiling liquid
- D. High boiling gas

17 The general formula for Alkene having one double bond is

- A. C_nH_{2n+1}
- B. C_nH_{2n}
- C. C_nH_{2n-2}
- D. C_nH_{2n+2}