

## Chemistry Fsc Part 2 Chapter 5 Online Test

Sr	Questions	Answers Choice
SI	Questions	A. an alkali metal
1	An element that has high ionization energy and tends to be chemically inactive would most likely to be	B. a transition element C. a noble gas D. a halogen
2	Which halogen occurs naturally in a positive oxidation state	A. Fluorine B. Chlorine C. Bromine D. lodine
3	Bleaching powder may be produced by passing chlorine over	A. calcium carbonate     B. hydrated calcium sulphate     C. calcium hydroxide     D. magnesium hydroxide
4	The anhydride of HClO₄is	A. CIO B. CIO <sub>2</sub> C. CIO <sub>3</sub> D. Cl <sub>2</sub> O <sub>7</sub>
5	Which halogen will react spontaneously with Au <sub>(s)</sub> to produce Au <sup>3+</sup>	A. Br <sub>2</sub> B. F <sub>2</sub> C. I <sub>2</sub> D. Cl <sub>2</sub>
6	Chlorine heptoxide (Cl <sub>2</sub> O <sub>7</sub> ) reacts with water to form	A. Hypochlorous acid B. Chloric acid C. Perchloric acid D. Chlorine and oxygen
7	Which of the following hydrogen halide is the weakest acid in solution	A. HF B. HBr C. HI D. HCI
8	Which of the following noble gas is used for arc welding and cutting	A. Helium B. Argon C. Xenon D. Radon
9	is use as a cooling medium for nuclear reactors	A. Ne B. He C. Ar D. Kr
10	Which one is chlorous acid	A. HCIO B. HCIO <sub>2</sub> C. HCIO <sub>3</sub> D. HCIO <sub>4</sub>
11	Which is the strongest acid	A. HCIO B. HCIO <sub>2</sub> C. HCIO <sub>3</sub> D. HCIO <sub>4</sub>
12	The anhydride of HClO₄is	A. CIO <sub>3</sub> B. CIO <sub>2</sub> C. CI <sub>2</sub> O <sub>5</sub> D. CI <sub>2</sub> O <sub>7</sub>
13	Which halogen will react spontaneously with Au(s) to produce Au <sup>3+</sup>	A. Br <sub>2</sub> B. F <sub>2</sub> C. I <sub>2</sub> D. Cl <sub>2</sub>
14	Hydrogen bond is the strongest between the molecules of	A. HF B. HCI C. HBr D. HI
15	Which one is perchloric acid	A. HCIO B. HCIO C. HCIO <sub>3</sub> D. HCIO <sub>4</sub>

6	Which is the second most abundant element in the universe	A. H B. He C. CO D. C
17	Which one of halogens is a liquid	A. F <sub>2</sub> B. Cl <sub>2</sub> C. Br <sub>2</sub> D.   <sub>2</sub>