

## Chemistry Fsc Part 2 Chapter 4 Online Test

| Sr | Questions  | Answers Choice  |
|----|--|---|
| 1  | Which of the following gives acidic oxide  | A. N<br>B. As<br>C. Sb<br>D. Bi   |
| 2  | Which of the following is a reddish brown gas                                      | A. $\text{N}_{2}\text{O}_{3}$<br>B. $\text{NO}_{2}$<br>C. $\text{N}_{2}\text{O}_{5}$<br>D. $\text{N}_{2}\text{O}_{7}$ |
| 3  | The oxidation of NO in air produces  | A. $\text{N}_{2}\text{O}_{3}$<br>B. $\text{NO}_{2}$<br>C. $\text{N}_{2}\text{O}_{5}$<br>D. $\text{N}_{2}\text{O}_{7}$ |
| 4  | The brown gas formed, when metal reduce $\text{HNO}_3$                             | A. NO<br>B. $\text{NO}_{2}$<br>C. $\text{N}_{2}\text{O}_{3}$<br>D. $\text{N}_{2}\text{O}_{5}$                         |
| 5  | Laughing gas is chemically   | A. NO<br>B. $\text{NO}_{2}$<br>C. $\text{N}_{2}\text{O}$<br>D. $\text{N}_{2}\text{O}_{4}$                             |
| 6  | Out of all the elements of group VA, the highest ionization energy is possessed by | A. N<br>B. As<br>C. Sb<br>D. Bi   |
| 7  | The most electronegative element of group V-A is                                   | A. N<br>B. P<br>C. Sb<br>D. Bi  |