

## Chemistry Fsc Part 2 Chapter 4 Online Test

Sr	Questions	Answers Choice
1	What are the number of the electrons in valence shell of P in PCl3	A. 4 B. 6 C. 8 D. 10
2	In aqua regia, the ratio of conc. HCl to Conc. HNO3 is	A. 1 : 1 B. 2 : 1 C. 1:2 D. 3 : 1
3	Which allotrope of phosphorus has layers like graphite.	<ul><li>A. white phosphorus</li><li>B. Red phosphorus</li><li>C. Black Phosphorus</li><li>D. Amorphous phosphorus</li></ul>
4	Basicity of ortho phosphoric acid is.	A. 1 B. 2 C. 3 D. 4
5	PCI3 reacts with water to form	A. PH3 B. POCI3 C. H3PO4 D. H3PO5
6	Each of the following is true for white and red phosphorus except one.	A. Both are soluble in CCl4     B. Both can be oxidized by heating in air     C. Both consists of same kind of atoms     D. Both can be converted into each other
7	P2O5 is usually used as	<ul><li>A. Drying agent only</li><li>B. Reducing agent</li><li>C. Both drying and reducing agent</li><li>D. Both drying agent and oxidizing agent.</li></ul>
8	Which form of phosphorus is more stable.	A. White B. Red C. Black D. Both a and b
9	Bone ash contain calcium phosphate	A. 40% B. 50% C. 70% D. 80%
10	Which raw material is used for manufacture eof HNO3 by Birkland eyed process	A. NH3 and CO2 B. Air C. Air and gypsum D. Lime stone and urea
11	Which of the following acids possess oxidizing and reducing properties.	A. HCI B. HNO2 C. HNO3 D. H2SO4
12	When Cu reacts with conc. HNO3 , which one of the following gases is evolved	A. N2O B. NO C. NO2 D. N2O5
		A. N2O B. NO
13	NH4NO3 on heating at 200 <sup>o</sup> C changes to	C. NO2 D. N2O4
14	NO2 can be obtained by heating.	A. NaNO3 B. KNO3 C. Pb(NO3)2 D. NH3NO3
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15	Which one of the following oxide is brown in colour.	A. NO B. NO2 C. N2O D. N2O3
16	Which element does not have allotropic form	A. Nitrogen B. Phosphorous C. Arsenic D. Antimony
17	Lowest oxidation state of nitrogen is present in.	A. NH3 B. NO2 C. NO D. HNO3