

Chemistry Fsc Part 2 Chapter 2 Online Test

Sr	Questions	Answers Choice
1	Which one is not use of lime	A. Extraction of metals B. Ceramic industry C. Preparation of bleaching powder D. Preparation of gypsum
2	When water is added to plaster of paris, it changes to a hard mass. Its volume also expands upto	A. 1% B. 2% C. 3% D. 5%
3	Sulphur is essential constituent for plants. Which is not role of sulphur	A. Chlorophyll development B. Development of root system C. Constituents of some proteins D. Increase transpiration force
4	Which of the following compounds is used to make spray, which has fungicidal action.	A. BaSO ₄ B. Na ₂ CO ₃ C. CaO D. CaSO ₄ .2H ₂ O
5	Which substance is used to lower the melting point of NaCl in Down's cell	A. CaCO ₃ B. CaCl ₂ C. MgSO ₄ D. AlCl ₃
6	In Down's cell anode is made up of.	A. Iron B. Copper C. Graphite D. Platinum
7	Sodium is collected over molten sodium chloride in Down's cell, because.	A. It does not react with NaCl B. It is lighter in weight and floats on molten NaCl C. Its reactivity is greater than NaCl D. None of these
8	Which is insoluble in water.	A. BeSO ₄ B. MgSO ₄ C. CaSO ₄ D. BaSO ₄
9	Which is called milk of magnesia.	A. MgCO ₃ B. Mg(OH) ₂ C. MgSO ₄ D. MgCl ₂
10	Which oxide is amphoteric in nature.	A. BeO B. MgO C. CaO D. BaO
11	Which one of the following alkaline earth metal form peroxide when heated with oxygen at 600 °C	A. Magnesium B. Calcium C. Strontium D. Barium
12	Which one of the following elements has the same oxidation number in all its known compounds.	A. Beryllium B. Nitrogen C. Sulphur D. Chlorine
13	Setting of Plaster of Paris is accompanied with	A. Hydrogen bonding B. Hydration C. Dehydration D. None of these
14	Which element reacts with alkali and H ₂ gas is produced.	A. Be B. Mg C. Ca D. Sr
15	Oxides of alkaline earth metal are	A. Acidic B. Basic C. Neutral

D. Amphoteric

16 Oxide of Alkaline earth metal are

- A. Acid
- B. Basic
- C. Neutral
- D. Amphoteric

17 When carbon di oxide is bubbled through lime water, white precipitate is formed This precipitate is.

- A. COCl_2
- B. H_2CO_3
- C. CaO
- D. CaCO_3