

Chemistry Fsc Part 2 Chapter 11 Online Test

Sr	Questions	Answers Choice
1	Which compound shows hydrogen bonding.	A. C2H6 B. C2H5CI C. CH3-O-CH3 D. C2H5OH
2	Ethanol reacts with Na metal to form sodium ethoxide. What product will be formed when C2H5ONa reacts with methyl bromide.	A. C2H5OC2H5 B. C2H5OCH3 C. CH3COC2H5 D. C2H5Br and NaBr
3	Ethanol on dehydration can be changed to	A. Ethene B. Diethyl ether C. Both 'a' and 'b' D. None of these
4	Ethanol on oxidation in the presence of K2Cr2O7 and Conc. H2SO4 changes to.	A. Acetaldehyde B. Ethane C. Ethene D. CO2 and H2O
5	Absolute alcohol is that which is	A. 100% B. 95% C. Ethanol mixed with methanol D. Ethanol mixed with H2O
6	The product of fermentation of sucrose is	A. Ethanol and H2O B. Ethanol and CO C. Ethanol and CO2 D. Glucose and CO2
7	According to Lewis concept, ethers behave as	A. Acid B. Base C. Nucleophile D. Solvent
8	Bakelite is obtained from phenol by reacting with	A. Acetal B. Ethanal C. Formaldehyde D. Methanol
9	The most reactive alcohol when O-H bond breaks is	A. Tertiary alcohol B. Secondary alcohol C. Primary alcohol D. Methyl alcohol
10	Isopropyl alcohol on oxidation gives	A. Acetaldehyde B. Acetone C. Ether D. Propene
11	Which compound is more soluble in water	A. C ₂ H ₅ OH B. C ₆ H ₅ OH C. CH ₃ COCH ₃ D. n - hexanol
12	Which one the following a dihydric alcohol	A. Ethanol B. Cyclo hexanol C. Glycerol D. Glycol
13	Ethanol can be converted into ethanoic acid by	A. Hydrogenation B. Hydration C. Oxidation D. Fermentation
14	Which compound show hydrogen bonding	A. C ₂ H ₆ B. C ₂ H ₅ Cl C. CH ₃ OCH ₃ D. C ₂ H ₅ OH
15	Which enzyme is not involved in fermentation of starch	A. Zymase B. Urease

		C. Invertase D. Diastase
16	compound shows extensive hydrogen bonding with water	A. C ₂ H ₆ B. H ₂ S C. C ₂ H ₅ OH D. CH ₃ Cl
17	Rectified spirit contains alcohol about	A. 80% B. 85% C. 90% D. 95%