

## Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Electron affinity is measure of energy	A. Required to remover the electron B. Released by adding an electron C. Required to excite an electron D. Released by removing an electron
2	Ionization energy of calcium is.	A. Lower than that of barium B. Lower than that of magnesium C. Higher than that of beryllium D. Lower than that of strontium
3	Hydrogen resembles in properties with	A. IA, IV A and VII A elements B. III A, IV A and V A elements C. II A, IV A and VI A elements D. II A, III A and VII A elements
4	Which statement is incorrect.	A. All the metals are good conductors of electricity B. All the metals are good conductor of heat C. All the metal form positive ions D. All the metal form acidic oxides
5	Mark the correct statement.	A. Na <sup>+</sup> is smaller than Na atom B. Na <sup>+</sup> is large than Na atom C. Cl <sup>-</sup> is smaller than Cl atom D. cl <sup>-</sup> and Cl are equal in size
6	Keeping in view size of atoms ,which order is the correct one.	A. Mg > Sr B. Ba > Mg C. Lu > Ce D. Cl > I
7	The element of 2nd period, which has highest ionization energy from the following is	A. Be B. C C. N D. O
8	Which of the following the highest hydration energy	A. Li <sup>+</sup> B. Na <sup>+</sup> C. K <sup>+</sup> D. Mg <sup>++</sup>
9	Which of the following has highest M.P	A. Aluminium B. Silicon C. Phosphorus D. Sulphur
10	Correct order according to atomic size in the following is	A. Na > K B. Be > Mg C. O > N D. Cl > F
11	Which the correct statement	A. Cl <sup>-</sup> is smaller than Cl atom B. Cl <sup>-</sup> (ion) and Cl (atom) are equal in size C. Na <sup>+</sup> is smaller than Na atom D. Na <sup>+</sup> is larger than Na atom
12	Which element has lowest melting point	A. Beryllium B. Magnesium C. Calcium D. Barium
13	Which of the following element has lowest ionization energy	A. Beryllium B. Boron C. Carbon D. Oxygen
14		A. Na atom is smaller than Na <sup>+</sup> B. Na atom is larger than K atom

14	Which of the following statement is correct	C. F atom is smaller than F <sup>-</sup> D. F atom is larger than F <sup>-</sup>
15	Which is the longest periodic table	A. 4 B. 5 C. 6 D. 7
16	Number of elements in the first period of the periodic table is	A. 2 B. 8 C. 14 D. 18
17	Which one is an incomplete period	A. 4th B. 5th C. 6th D. 7th