

Chemistry Fsc Part 2 Chapter 1 Online Test

Sr	Questions	Answers Choice
1	Which one of the following is a metalloid.	A. sulphur B. Antimony C. Mercury D. Zinc
2	Which one of the following elements has the largest second ionization energy.	A. O B. Na C. F D. Ne
3	Which class of elements shows low value of first ionization potential.	A. Alkali metals B. Alkaline earth metals C. Halogens D. Noble gases
4	As you proceed across a period in the periodic table the first ionization energy	A. Decrease B. Increase C. Remains constant D. First increase up the middle of period and then decreases
5	In a period, from left to right in the periodic table, the size of atom generally.	A. Increases B. decreases C. Remains constant D. First increase upto the middle of period and then decreases
6	In which one of the following sets do all three particles have same number of total electrons	A. F ⁻ , Cl ⁻ , Br ⁻ B. Li ⁺ , Na ⁺ , K ⁺ C. N ³⁻ , O ²⁻ , F ⁻ D. Na ⁺ , Mg ²⁺ , K ⁺
7	In which block of periodic table non metals are present.	A. s B. p C. d D. f
8	Transition elements are those	A. Which comes before uranium B. Which comes after uranium C. Which are prepared artificially D. Which are in between s-block and p-block elements
9	Majority of the elements of the periodic table are.	A. Semi metals B. Non metals C. Metals D. Noble metals
10	What criteria did Mendeleev use in arranging his periodic table.	A. Atomic mass B. Atomic number C. Mass number D. Density
11	What are the total numbers of periods in the modern periodic table.	A. 3 B. 5 C. 7 D. 8
12	What are the total numbers of periods in the modern periodic table	A. 3 B. 5 C. 7 D. 8
13	Fluorine is in group VIIA of periodic table. Its chemistry will most closely resemble that of.	A. Argon B. Boron C. Iodine D. Sulphur
14	Fluorine is in group VII A of periodic table. Its chemistry will most closely resemble that of.	A. Argon B. Boron C. Iodine D. Sulphur
		A. Cu, Hg, Au

15	Which one of the following sets has coinage metals is it.	<div>B. Cu, Ag, Au</div> <div>C. Ag, Au, Hg</div> <div>D. Cu, Fe, Au</div>
16	In which group of periodic table is the element which has atomic number 14.	<div>A. II</div> <div>B. IV</div> <div>C. III</div> <div>D. VI</div>
17	Mark the incorrect statement	<div>A. Metallic character increase down the group</div> <div>B. Metallic character increase from left to right along a period</div> <div>C. Metallic character decrease from left to right along a period</div> <div>D. Metallic character remains the same down the group</div>