

Chemistry 9th Class English Medium Unit 7 Online Test

Sr	Questions	Answers Choice
1	Which one is not an Arrhenious acid?	A. HCl B. H ₂ SO ₄ C. CO ₂ D. HNO ₃
2	Which one is not an Arrhenius base.	A. KOH B. NaOH C. NH ₃ D. Ca(OH) ₂
3	According to Arrhenius concept base is a substance which dissociates in aqueous solution to give	A. Hydroxide ions B. Hydrogen ions C. Pair of Electrons D. Proton
4	According to Arrhenius concept acid is a substance which dissociates in aqueous solution to give.	A. Proton B. Pair of Electron C. Hydrogen ions D. Hydroxide ion
5	Milk of magnesia contains Mg(OH) ₂ It is used as antacid It neutralizes excess stomach acid. Which salt is formed in this reaction.	A. MgCl ₂ B. MgSO ₄ C. MgCO ₃ D. MgO
6	Which of the following cannot be classified as Arrhenius acid	A. CO ₂ B. HNO ₃ C. H ₂ CO ₃ D. H ₂ SO ₄
7	All bases turn red litmus.	A. Blue B. White C. Pink D. Colourless
8	All Acids turn blue litmus.	A. Pink B. White C. Colourless D. Red
9	Acids means	A. Bitter B. Salty C. Sweet D. Sour
10	Acidic Acid is used for	A. Cleaning metal B. Etching designs C. Flavouring food D. Making explosives
11	A base is a substance which neutralizes an acid. Which of these substances is not a base.	A. Aqueous ammonia B. Calcium oxide C. sodium Chloride D. Sodium carbonate
12	When a chemical reaction is carried out with a substance Z; a gas is produced which turns red litmus paper blue. What is the reaction?	A. Reaction of an acid with a metal carbonate B. Reaction of an alkali with ammonium salt C. Reaction of an acid with ammonium salt D. None of these