

Chemistry 9th Class English Medium Unit 3 Online Test

Sr	Questions	Answers Choice
1	Which metal has the lowest melting point?	A. Li B. Na C. Rb D. K
2	Keeping in view the intermolecular forces of attraction, indicate which compound has the highest boiling point	A. H_2S B. HF C. NH_3 D. H_2O
3	Which form of carbon is used as a lubricant?	A. Coal B. Diamond C. Charcoal D. Graphite
4	Which of the following bond s is expected to b the weakest.	A. $Cl-Cl$ B. $C-C$ C. $F-F$ D. $O-O$
5	Why is H_2O liquid while H_2S is a gas?	A. Becuase in waer, the atomic size of oxygen in smaller than that of Sulphur B. Because water can easily freeze into ice C. Because water is a polar compound ad tere exists strong forces of attrction betwene its molecules D. Because H_2O molecule is lighter than H_2S
6	Which element is capable of forming all the three types of bonds, covalent coordinate covalent or ionic.	A. Carbon B. Silicon C. Magnesium D. Oxygen
7	When molten copper and olten zinc are mixed together , they give rise to a new substnce called brass. Predict what type of bond is formed between copper and zinc.	A. Ionic bond B. Coordinate Covalent bond C. Metallic bond D. Covalent Bond