

Chemistry 9th Class English Medium Unit 3 Online Test

Sr	Questions	Answers Choice
1	Which one of the following is the weakest force among the atoms.	A. Intermolecular force B. Ionic force C. Metallic force D. Covalent forces
2	Identify the compound which is not soluble in water	A. KBr B. $MgCl_2$ C. C_6H_6 D. NaCl
3	Triple covalent bond involves how many electrons.	A. Six B. Four C. Eight D. Three
4	How many covalent bonds does C_2H_2 molecule have.	A. Two B. Three C. Four D. Five
5	Covalent Bond involves the	A. Sharing of electrons B. Repulsion of electrons C. Acceptance of electrons D. Donation of electrons
6	A bond pair in covalent molecules usually has.	A. One electron B. Two electron C. Three electron D. Four electron
7	A bond formed between two non metals is expected to be	A. Ionic B. Coordinate covalent C. Metallic D. Covalent
8	Identify the covalent compound	A. NaCl B. H_2O C. KF D. MgO
9	In the formation of AlF_3 , aluminum atom loses.....electrons.	A. 1 B. 4 C. 3 D. 2
10	Silicon belongs to Group IVA . It haselectrons in the valence shell	A. 2 B. 6 C. 3 D. 4
11	How many covalent bonds do N_2 molecule have	A. 3 B. 4 C. 2 D. 5
12	Number of electrons in nitrogen molecule is.	A. 2 B. 4 C. 6 D. 8