

## Chemistry 9th Class English Medium Unit 3 Online Test

Sr	Questions	Answers Choice
1	None polar compounds are insoluble in	A. Alcohol B. Benzene C. Ether D. Water
2	Ionic compound have	A. Low melting and high boiling points B. Low melting and boiling point C. High melting and boiling points D. High melting and low boiling points
3	Ionic compound are good conductors electricity in	A. Solution B. Molten state C. Solid state D. both a and b
4	The compound formed by opposite charges are known as.	A. Metallic solids B. Ionic compounds C. Non-polar Covalent compound D. None of the above
5	Water has high boiling point as compared to alcohol due to	A. Low density B. High surface tension C. Hydrogen bonding D. High vapour pressure
6	The boiling point of alcohol is	A. $44^{\circ}\text{C}$ B. $78^{\circ}\text{C}$ C. $53^{\circ}\text{C}$ D. $19^{\circ}\text{C}$
7	The boiling point of water is	A. $0^{\circ}\text{C}$ B. $100^{\circ}\text{C}$ C. $35^{\circ}\text{C}$ D. $25^{\circ}\text{C}$
8	The force of attraction between water molecules is.	A. Ionic bonding B. Covalent bonding C. Hydrogen Bonding D. Coordinate Covalent bonding
9	Which of the following is an example of polar covalent compound.	A. $\text{Cl}_2$ B. $\text{H}_2$ C. $\text{O}_2$ D. $\text{HCl}$
10	Hydrogen bonding is always found in	A. Non-polar molecules B. Homoatomic molecules C. Polar Molecules D. All of the above
11	Metals have the tendency to lose electrons due to.	A. High ionization energies B. Low ionization energies C. Low electron affinity D. None of the above
12	Malleability is the property by virtue of which a metal can be drawn into.	A. Rods B. Plates C. Sheets D. Wires