

Chemistry 9th Class English Medium Unit 2 Online Test

Sr	Questions	Answers Choice
1	The nucleus of an atom is composed of	A. Electrons B. Electons and protons C. Protons and neutrons D. Electrons and neutrons
2	Proton are reflected toward plate.	A. Positive B. Negative C. Both a and b D. None of these
3	Quantum means.	A. Virable energy B. Fixed energy C. High energy D. Minimum energy
4	According to Rutherford's atomic theory, atom should produce.	A. Line spectrum B. Continuous spectrum C. Both a and b D. None of these
5	Who performed fist experiment to split atom	A. Bohr B. Newton C. Rutherford D. Soddy
6	Rutherford used a gold foil in his experiment, which had a tickness of	A. 0,0002 cm B. 0,0001 cm C. 0.001 cm D. 0.00004cm
7	John Dalton put forward his atomic theory.	A. 1800 B. 1803 C. 1805 D. 1903
8	A sub sheel that cna accommodate 6 electrons is	A. b B. s C. f D. p
9	M shell has sub sheels.	A. 1s , 2s B. 1s,2s,3s C. 2s,2p D. 3s,3p,3d
10	How doe electrons keep themselves away from the oppositely charged nucleus.	A. A agnetic field around the nucleus keeps them away B. By keeping themselves stationary C. By revolving around the nucleus D. Due to their wave like nature
11	What does keep the particles present in the nucleus intact.	A. Particles are held together by dipolar force B. Particles are held together by wee nuclear force C. Particles are held together by strong nuclear force D. Particles are held together by electrostatic force
12	How is radio carbon dating useful for archeolgists.	A. It helps detemine whether the matter is radioactie or not B. It helps determine the age of organic matter C. It helps determine the composition of matter D. It helps deterime the usefulness of matter