

## Chemistry 10th Class English Medium Unit 10 Online Test

Sr	Questions	Answers Choice
1	A conjugate base is a specie formed by donating a.	A. Proton B. Electron pair C. Neutron D. Electron
2	According to Bronsted and Lowry concept a base is a substance that can accept.	A. Proton B. Electron pair C. Neutron D. Electron
3	A conjugate acid is a specie formed by accepting a.	A. Proton B. Electron pair C. Neutron D. Electron
4	According to Bronsted and Lowry concept an acid is a substance that can donate.	A. Proton B. Electron pair C. Neutron D. Electron
5	Bronsted and Lowry presented their theories of acids and bases in	A. 1785 B. 1787 C. 1923 D. 1925
6	Which one is not an Arrhenius base?	A. NaOH B. KOH C. $\text{Ca}(\text{OH})_2$ D. $\text{NH}_3$
7	Which one is not an Arrhenius acid?	A. HCl B. $\text{H}_2\text{SO}_4$ C. $\text{CO}_2$ D. $\text{HNO}_3$
8	According to Arrhenius concept acid is a substance which dissociates in aqueous solution to give.	A. Hydrogen ions B. Hydroxide ions C. Both a and b D. None of these
9	Arrhenius presented his concept about acids and bases in.	A. 1785 B. 1787 C. 1923 D. 1930
10	All bases turn red litmus	A. Red B. Blue C. Pink D. White
11	All acids turn blue litmus.	A. Red B. Blue C. Pink D. White
12	Which acid is present in our stomach.	A. Nitric acid B. Hydrochloric acid C. Sulphuric acid D. All of these