

## Chemistry 10th Class English Medium Unit 9 Online Test

Sr	Questions	Answers Choice
1	When the numbers of moles of both sides are equal in a reaction, then the unit of $K_C$ will be:	<p>A. No unit</p> <p>B. <math>\text{mol}^{-2}</math></p> <p>C. <math>\text{dm}^6</math></p> <p>D. <math>\text{mol}^{-2} \text{dm}^3</math></p>
2	The reaction in which the number of moles reactants and products not equal in balance chemical equation the units of $K_C$ for this reactions are	<p>A. <math>\text{mol}^{-2}</math></p> <p>B. <math>\text{dm}^3</math></p> <p>C. <math>\text{mol}^2</math></p> <p>D. <math>\text{mol dm}^{-1}</math></p>
3	The value of $K_C$ depends on.	<p>A. Temperature</p> <p>B. Pressure</p> <p>C. Volume</p> <p>D. Atmosphere</p>
4	Formation process of Ammonia by the combination Hydrogen and Nitrogen was given by:	<p>A. Dalton</p> <p>B. Thomson</p> <p>C. Haber</p> <p>D. Waage</p>
5	The unit of molar concentration is:	<p>A. <math>\text{mol dm}^{-3}</math></p> <p>B. <math>\text{mol dm}^{+3}</math></p> <p>C. <math>\text{mol cm}^{-3}</math></p> <p>D. <math>\text{mol cm}^{+3}</math></p>
6	$K_C$ is always equal to	<p>A. <math>R_f/R_r</math></p> <p>B. <math>K_f /K_t</math></p> <p>C. <math>K_f/K_r</math></p> <p>D. <math>R_r/R_t</math></p>
7	Guldberg and waage out forward law of mass action in:	<p>A. 1889</p> <p>B. 1879</p> <p>C. 1869</p> <p>D. 1859</p>
8	Who proposed "Law of mass action"?	<p>A. Newton</p> <p>B. Boyle</p> <p>C. Guldberg and waage</p> <p>D. Lavoisier</p>
9	In the beginning the rate of reverse reaction is	<p>A. Negligible</p> <p>B. Moderate</p> <p>C. Very fast</p> <p>D. Slow</p>
10	In chemical reaction , the substances that combine are called.	<p>A. Reactants</p> <p>B. Products</p> <p>C. Equilibrium</p> <p>D. Numerator</p>
11	Which colour of HI is ?	<p>A. Orange</p> <p>B. Purple</p> <p>C. Red</p> <p>D. Colourless</p>
12	Such reactions which continue in both directions are called.	<p>A. Irreversible</p> <p>B. Reversible</p> <p>C. Nonreactive</p> <p>D. Dynamic</p>