

## ECAT Pre General Science Online Test

Sr	Questions	Answers Choice
1	When a shell explodes in mid-air, its fragments fly off in	<ul><li>A. only one direction</li><li>B. in two direction</li><li>C. different directions</li><li>D. a particular direction</li></ul>
2	Suppose the water flows out from a pipe at 3kg s <sup>-1</sup> and its velocity changes from 5m s <sup>-1</sup> to zero on striking the wall, then the force exerted by water on wall will be	A. 5 N B. 10 N C. 15 N D. 20 N
3	A snooker ball moving with velocity V collides head on with another snooker ball of same mass at rest. If the collision is elastic, the velocity of second snooker ball is	A. Zero B. Infinity C. V D. 2 V
4	An alpha particle has a charge of	A. +2e B2e Ce D. +3e
5	When a nucleus emits an alpha particles, its charge number decreases by	A. 3 B. 2 C. 6 D. 5
6	When a nucleus emits an alpha particle, it atomic mass decreased by	A. 2 B. 1 C. 4 D. 3
7	Radioactivity is	A. self disruptive activity B. spontaneous activity C. exhibited by all elements under proper conditions D. both 'a' and 'b'
8	Curie is a unit of	A. reluctance B. resistivity C. binding energy D. radioactivity
9	Alfa , beta and gamma rays are emitted from a radio-active substance	A. spontaneously B. when it is heated C. when it is exposed to light D. When it interacts with the other particle
10	Gamma rays consist of steam of	A. electron B. proton C. photons D. all of these
11	Alfa particles are	A. hydrogen nuclei B. helium nuclei C. electrons D. photons
12	Beta particles are	A. hydrogen nuclei B. helium nuclei C. electrons D. photons
13	Maric Curie and Pieree Curie discovered two new radioactive elements, which are called	A. polonium uranium B. uranium and radium C. polonium and radium D. none of these
14	Radioactivity was discovered by	A. Rutherford B. Henri Becqureal C. Maxwell D. James Chadwick
		A. is exhibited more by semiconductors in general B. in exhibited more by the element

15	Radioactivity	when they are coupled C. with other radioactive elements by a covalent bond D. is an atomic property of radioactive elements
16	Binding energy per nucleus is	A. greater for heavy nucleus     B. least for heavy nucleus     C. greatest for light nuclei     D. decreases for medium weight niclei
17	The amount of energy equivalent to 1 a.m.u is	A. 9.315 Mev B. 93.15 Mev C. 931.5 Mev D. 2.22 Mev
18	The energy is found from Einstein's mass energy relation is called	A. binding energy of electron B. binding energy of proton C. binding energy of neutron D. binding energy of nucleus
19	The missing mass which is converted to energy in the formation of nucleus, is called	A. packing fraction B. mass defect C. binding energy D. none of these
20	The energy acquired by a mass of 1g moving with the speed of light is	A. 3 x 10 <sup>8</sup> J B. 9 x 10 <sup>13</sup> J C. 3 x 10 <sup>13</sup> J D. 9 x 10 <sup>16</sup> J
21	If 'V' is the relativistic speed and 'C' is the speed of light then according to Einstien the factor V/C must always be	A. Equal to 1 B. Less than 1 C. Greater than 1 D. Infinity
22	1 amu is equal to.	A. 1.66 x 10 <sup>-24</sup> kg B. 1.66 x 10 <sup>-19</sup> kg C. 1.66 x 10 <sup>-24</sup> kg D. 1.66 x 10 <sup>-24</sup> kg
23	The mass of the nucleus is always less than the total man of the protons and neutron that make up the nucleus. The difference of the two masses is called	A. nuclear fission B. nuclear fusion C. man defect D. radioactivity
24	Neon gas have three isotopes whose atomic numbers are	A. 20, 24, 23 B. 20, 21, 22 C. 20, 19, 21 D. none of these
25	The most abundant isotope of neon is	A. neon-20 B. neon-21 C. neon-22 D. neon-23
26	A mass spectrograph sort out	A. molecules B. atoms C. elements D. isotopes
27	The chemical properties of an element depends upon the number of	A. electron B. position C. photons D. neutrons
28	The chemical properties of all the isotopes of an elements are	A. same B. different C. slightly different D. none of these
29	Hydrogen atom with only one proton and one neutron in its nucleus, and one electron, is called	A. deuterium B. protium C. tritium D. none of these
30	Hydrogen atom with only one proton in its nucleus, and one electron in its orbit is called	A. deuteron B. deterium C. protium D. tritium