

## ECAT Physics Online Test

Sr	Questions	Answers Choice
1	Each atom in metal crystal:	A. Remains fixed B. Vibrates about a fixed point C. Moves randomly D. Rotates about center of a crystal E. None of these
2	The smallest three dimensional basic structure is called as:	A. An atom B. Unit cell C. Crystal lattice D. Polymer E. None of these
3	In crystalline solids, atoms are held about their equilibrium positions depending upon the strength of:	A. Adhesive force B. Nuclear forces C. Inter atomic cohesive force D. Electromagnetic force E. None of these
4	The pattern of NaCl particles have a shape which is :	A. Cubic B. Body centred cubic C. Simple cubic D. face centred E. Both (A) and (C)
5	The whole structure obtained by the repetition of unit cells is called:	A. Crystal lattice B. Amorphous solid C. Polymeric solid D. Polyesterne E. None of these
6	The temperature at which the vibrations become so great that structure of the Crystal breaks up, is called:	A. Critical temperature B. Temperature of vaporization C. Melting point D. Both (A) and (C) E. Both (A) and (B)
7	Tick the one which is not a crystalline solid:	A. Zirconia B. Glass C. Copper D. Ceramic solid E. An ionic compound
8	A unit cell is smallest basic structure which is:	A. One dimensional B. Two dimensional C. Three dimensional D. Four dimensional E. None of these
9	The arrangement or molecules or atoms in a crystalline solid can be studied by using:	A. Chemical methods B. Neutrons C. X-ray techniques D. Copper atoms E. Both (A) and (B)
10	An ordinary glass gradually softens into a 'paste -like' state before it becomes a very viscous liquid. It happens almost at:	A. 800 <sup>o</sup> C B. 500 <sup>o</sup> C C. 300 <sup>o</sup> C D. 100 <sup>o</sup> C E. None of these
11	In a cubic crystal, All solids meet at:	A. 60 <sup>o</sup> B. 90 <sup>o</sup> C. 109 <sup>o</sup> D. 30 <sup>o</sup> E. 10 <sup>o</sup>
12	The pattern of crystalline solid is:	A. One dimensional B. Two dimensional C. Three dimensional D. None of these E. Either (A) or (B)
13	Amorphous solids:	A. Have definite melting points B. Are called glassy solids C. Have no definite melting point D. Both (A) and (B)

		<p>D. Both (B) and (C)</p> <p>E. Both (A) and (C)</p>
14	The word amorphous means:	<p>A. Without any structure</p> <p>B. With definite structure</p> <p>C. Regular arrangement of molecules</p> <p>D. Both (B) and (C)</p> <p>E. None of these</p>
15	The force which maintain the strict long-range order between atoms of a crystalline solid is the:	<p>A. Nuclear force</p> <p>B. Cohesive force</p> <p>C. Adhesive force</p> <p>D. Coulomb force</p> <p>E. None of these</p>
16	The transition from solid state to liquid state is:	<p>A. Abrupt</p> <p>B. Slow</p> <p>C. Continous</p> <p>D. Discontinuous</p> <p>E. Both (A) and (D)</p>
17	The transition from solid to liquid is actually from:	<p>A. Order to disorder</p> <p>B. Disorder to order</p> <p>C. Order to order</p> <p>D. Disorder to disorder</p> <p>E. None of these</p>
18	Each atom in a metal crystal vibrates about a fixed point with an amplitude that:	<p>A. Decrease the rise in temprature</p> <p>B. Is not affected by rise in temprature</p> <p>C. Increase with rise in temprature</p> <p>D. Both (B) and (C)</p> <p>E. None of these</p>
19	Zirconia is classified as:	<p>A. Ceramic solid</p> <p>B. Ionic compound</p> <p>C. Metal</p> <p>D. Either (A) or (B)</p> <p>E. Either (B) or (C)</p>
20	The solids are classified as:	<p>A. Metals</p> <p>B. Crystalline</p> <p>C. Amorphous</p> <p>D. Polymeric</p> <p>E. All except (A)</p>
21	Crystalline solids are in the form of:	<p>A. Metals</p> <p>B. Ionic Compounds</p> <p>C. Ceramics</p> <p>D. Both (A) and (B)</p> <p>E. All of these</p>
22	The basic circuit element in D.C. circuit is:	<p>A. A capacitor</p> <p>B. A resistor</p> <p>C. An inductor</p> <p>D. Both (A) and (C)</p> <p>E. Both (A) and (B)</p>
23	The basic circuit element in A.C. circuits are:	<p>A. Resistor and capacitor</p> <p>B. Resistor and Inductor</p> <p>C. Capacitor only</p> <p>D. Both (B) and (C)</p> <p>E. None of these</p>
24	Unless stated otherwise, when we speak of A.C. meter reading, we usually mean:	<p>A. Peak value</p> <p>B. RMS value</p> <p>C. Instantaneous value</p> <p>D. Peak-to-peak value</p> <p>E. Both (A) and (C)</p>
25	The length of rotating vector (on a certain scale) represents the:	<p>A. Peak value of alternating quantity</p> <p>B. RMS value of alternating quantity</p> <p>C. Instantaneous value of alternating quantity</p> <p>D. Either (B) or (C)</p> <p>E. Either (A) or (B)</p>
26	A sinusoidally alternating voltage or current can be graphically represented by a:	<p>A. Vector</p> <p>B. Rotating vector</p> <p>C. Clockwise vector</p> <p>D. Anticlockwise voltage vector</p> <p>E. None of these</p>
27	If 250V is the RMS value of alternative voltage, then its peak value $V_0$ will be:	<p>A. 353.5V</p> <p>B. 250V</p> <p>C. 175V</p> <p>D. zero</p> <p>E. 400V</p>

A. RMS value

28 If we connect a A.C. volt meter to read A.C. voltage, It would read its:

- A. RMS value  
B. Instantaneous value  
C. Valued average over a cycle  
D. Zero  
E. Both (B) and (C)

29 The phase at the positive peak of an A.C. cycle is:

- A. 0°<span style=
- B. 90°<span style=
- C. 180°<span style=
- D. 0 and 2>>