

ECAT Chemistry Online Test

Sr	Questions	Answers Choice
1	Hydrolysis of Grignard's reagent gives:	A. Alcohol B. Halide C. Alkyl D. Alkane
2	Organic compounds containing halogen atom are called:	A. R---OH B. R----X C. R---NH ₂ D. R---COH
3	The order of reactivity for a given halogen in Grignard's reagent is:	A. CH ₃ >X > C ₂ H ₅ >X > C ₃ H ₇ >X > C ₄ H ₉ >X B. C ₂ H ₅ >X > C ₃ H ₇ >X > C ₄ H ₉ >X > CH ₃ >X C. C ₃ H ₇ >X > C ₄ H ₉ >X > CH ₃ >X > C ₂ H ₅ >X D. C ₄ H ₉ >X > C ₃ H ₇ >X > C ₂ H ₅ >X > CH ₃ >X
4	Reactivity of alkyl halides with magnesium is of the order:	A. RI > RBr > RCl > RF B. RBr > RCl > RF > RI C. RCl > RF > RI > RBr D. RF > RI > RBr > RCl
5	Grignard's reagent was prepared in:	A. 1900 B. 1910 C. 1920 D. 1930
6	Steps in SN ₂ reactions are:	A. One B. Two C. Three D. Four
7	Electronegativity order of alkyl halides is:	A. RI > RBr > RCl > RF B. RBr > RCl > RF > RI C. RCl > RF > RI > RBr D. RF > RI > RBr > RCl
8	Catalyst in the reaction ROH + SOCl ₂ → RCl + SO ₂ + HCl is:	A. ZnCl ₂ B. Pyridine C. H ₂ SO ₄ D. Ether
9	Primary carbon attaches with other hydrogen atoms directly:	A. One B. Two C. Three D. At least one or more than it
10	Reaction of which with Grignard's reagent gives primary alcohol:	A. Formaldehyde B. Aldehyde C. Ketones D. Acetone
11	Reaction of Grignard's reagent with CO ₂ gives:	A. Aldehyde B. Pri-alcohol C. Sec-alcohol D. Carboxylic acid
12	Metal used in the preparation of Grignard's reagent is:	A. Ca B. Na C. Mg D. Zn
13	E ₂ has molecularity :	A. One B. Two C. Three D. Four

		D. Hair
14	SN ₂ reaction has order of reaction :	A. First B. Second C. Third D. Zero
15	Alkyl halides are reactive :	A. High B. Medium C. Less D. Least
16	Best method of preparation of alkyl halide from alcohols is by its reaction with:	A. HX B. SOCl ₂ C. P ₅ and P ₃ D. All
17	General formula of alkyl halide is:	A. R-X B. R-OH C. R-COOH D. R-COOH
18	which one of the following is not a nucleophile?	A. H ₂ O B. H ₂ S C. BF ₃ D. NH ₃
19	The rate of E ₁ reaction depends upon:	A. The concentration of substrate B. The concentration of nucleophile C. The concentration of substrate as well as nucleophile D. None of these
20	Alkyl halides are considered to be very reactive compounds towards nucleophiles, because:	A. They have an electrophilic carbon B. They have an electrophilic carbon and good leaving group C. They have an electrophilic carbon and bad leaving group D. They have an nucleophilic carbon and good leaving group
21	For which mechanisms, the first step involved is the same:	A. E ₁ and E ₂ B. E ₂ and SN ₂ C. E ₂ and E ₁ D. E ₁ and SN ₁
22	Elimination bimolecular reactions involve:	A. First order Kinetics B. Third order kinetics C. Zero order kinetics
23	SN ₂ reactions can be best carried out with:	A. Pri. alkyl halide B. Sec. Alkyl halide C. Ter. Alkyl halide D. All of three
24	Grignard reagent is reactive due to :	A. The presence of halogen atom B. The presence of Mg atom C. The polarity of C - Mg bond D. None of them
25	When CO ₂ is made to react with ethyl magnesium iodide, followed by hydrolysis, the product formed is:	A. Propane B. Propanoic acid C. Propanal D. Propanol
26	The reactivity order of alkyl halides for a particular alkyl group is:	A. F > Cl > Br > I B. Cl > Br > F > I C. I > Br > Cl > F D. Br > I > Cl > F
27	In primary alkyl halides, the halogen atom is attached to a carbon which attached to how many carbon atoms?	A. Two B. Three C. One D. Four
28	Hydrocarbons contain :	A. Carbon only carbon B. Hydrogen only C. Carbon & hydrogen D. Carbon , hydrogen & halogen
		A. CH ₃ -C B. CH ₃ -C C. CH ₃ -C D. CH ₃ -C

29 Which one gives acidic reactions?

- ~~A. $\text{CH}_3\text{CH}_2\text{CH}_3$~~
B. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3$
 $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$
C. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$
 $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$
D. $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_3$

30 Mustard gas is a :

- A. Gas
B. Liquid
C. Solid
D. High boiling point