

ECAT Chemistry Chapter 15 Group IV-A and VI-A Elements Online Test

Sr	Questions	Answers Choice
1	What are non-metals?	A. Sb & Bi B. P & Bi C. N & P D. As & N
2	Which element belongs to group V-A?	A. Cs B. Ba C. Sr D. Bi
3	Which gas is heavier than air?	A. CO ₂ B. NH ₃ C. Cl ₂ D. CO ₂
4	Which has sweetish taste?	A. N ₂ O B. NH ₃ C. Cl ₂ D. CO ₂
5	Which is nitrous acid?	A. HNO ₃ B. HNO C. HNO ₂ D. H ₂ NO ₃
6	Formation of NO from N_2 and O_2 requires temp	A. 30 ⁰ C B. 300 ⁰ C C. 3000 ⁰ C D. 30000 ⁰ C
7	Which can be prepared from HNO ₃ ?	A. T.N.T B. Fertilizers C. Picric acid D. All of these
8	Which is laughing gas?	A. NO B. N ₂ O C. NO ₂ D. NO ₃
9	Which is aqua regia?	A. HCL: NO B. HCl: HNO ₃ C. HCl: H ₂ SO ₄ D. H ₂ SO ₄ : HNO ₃
10	Sulphuric acid is used	A. As fertilizersB. As dehydrating agentC. As oxidizing agentD. All above
11	Calcium carbonate contains oxygen in it	A. 48% B. 50% C. 53% D. 60%
12	Which are metalloid?	A. Nitrogen and phosphorous B. Arsenic and phosphorous antimony C. Phosphorous and arsenic D. Antimony and bismuth
13	Nitrogen is present in atmosphere by weight	A. 75% B. 76% C. 77% D. 78%
14	Which phosphorus is the most poisonous?	A. White B. Red C. Black D. All
15	Aqua Regia has ratio of conc. HCl and \ensuremath{HNO}_3	A. 1 : 2 B. 1 : 3 C. 1 : 4

		D. 2:3
16	NO ₂ forms acidic solution	A. Red B. Blue C. Green D. Brown
17	Tungsten and uranium are turned to an reaction with \mbox{HNO}_3	A. Oxides B. Chlorides C. Nitrides D. Nitrates
18	Reaction of HNO, with very dilute Zinc and at low temperature gives	A. H ₂ B. NO C. NO ₂ D. NH ₄
19	Which has garlic like odour?	A. N ₂ O B. NO C. NO ₂ D. NH ₄
20	Which of the following species has the maximum number of unpaired electrons?	A. O ₂ B. O ₂ ⁺ C. O ₂ ⁻ D. O ₂ ⁻²
21	Which catalyst is used in contact process?	A. Fe ₂ O ₃ B. V ₂ O ₅ C. SO ₃ D. Ag ₂ O
22	SO ₃ is not absorbed in water directly to or H ₂ SO ₄ because	A. the reaction does not go to completion. B. the reaction is quite slow. C. the reaction is exothermic. D. SO ₃ is insoluble in water.
23	Out of all the elements of group V1-A the highest melting and boiling points is shown by the element	A. Te B. Se C. S D. Pb
24	Laughing gas is chemically	A. NO B. N _{2O} C. NO ₂ D. N ₂ O ₄
25	The brown gas formed when metal reduces HNO3is	A. N ₂ O ₅ B. N ₂ O ₃ C. NO ₂ D. NO
26	Oxidation of NO in air produces	A. N ₂ 0 B. N ₂ O ₃ C. N ₂ O ₄ D. N ₂ O ₅
27	Among group V-A elements, the most electronegative element is	A. Sb B. N C. P D. As
	Out of all the elements of group V-A, the highest ionization energy is possessed by	A. N B. P C. Sb