

ECAT Chemistry Chapter 12 Periodic Classification of Elements and Periodicity Online Test

Sr	Questions	Answers Choice
1	The electropositive elements from	A. Acidic oxides B. Basic oxides C. Neutral oxides D. None
2	According to Mendeleev, the physical and the chemical properties are the periodic function of their	A. Atomic number B. Atomic mass C. Atomic wt D. None
3	Transition elements have valence electrons in	A. s-orbital B. p-orbital C. d-orbital D. f-orbital
4	Rare earth elements are	A. s-block elements B. p-block elements C. d-block elements D. f-block elements
5	The IA elements are called	A. Alkaline earth metal B. Alkaline metals C. The halogens D. The inert gases
6	The number of elements in the first, second and third period are	A. 2, 8, 18 B. 8, 2, 18 C. 2, 18, 8 D. 2, 8, 8
7	The horizontal rows in the periodic table are called periods. The number of period are	A. 5 B. 6 C. 7 D. 8
8	The number of groups in the periodic table is	A. 6 B. 7 C. 8 D. 9
9	The period table contains elements in vertical column. these vertical column are called	A. Groups B. Periods C. Blocks D. Sub group
10	In the modern periodic table the elements are placed in the ascending order of their	A. Atomic masses B. Melting points C. Boiling points D. Atomic numbers
11	Doberiner arranged the similar elements into	A. Pairs B. Triads C. Triplets D. Rows