

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Which is the least reactive of all the alkali metals	A. Li B. Na C. K D. Cs
2	The mineral $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ has the general name	A. Dolomite B. Gypsum C. Calcite D. Epsom Salt
3	Dolomite is a carbonate of	A. Be B. Mg C. Na D. Ba
4	----- does not belong to Alkaline-Earth metals	A. Be B. Ra C. Ba D. Rn
5	Chile saltpeter has the chemical formal	A. NaNO_3 B. KNO_2 C. $\text{Na}_2\text{B}_4\text{O}_7$ D. $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$
6	Which one of the following is not an alkali metal	A. Francium B. Caesium C. Rubidium D. Radium
7	Which one does not belong to the alkaline earth metals	A. Be B. Ba C. Ra D. Rn
8	The word alkali is derived from which language	A. Arabic B. Greek C. French D. German
9	Which of the following the highest hydration energy	A. Li^+ B. Na^+ C. K^+ D. Mg^{2+}
10	Which of the following has highest M.P	A. Aluminium B. Silicon C. Phosphorus D. Sulphur
11	Correct order according to atomic size in the following is	A. $\text{Na} > \text{K}$ B. $\text{Be} > \text{Mg}$ C. $\text{O} > \text{N}$ D. $\text{Cl} > \text{F}$
12	Which the correct statement	A. Cl^- is smaller than Cl atom B. Cl^- (ion) and Cl (atom) are equal in size C. Na^+ is smaller than Na atom D. Na^+ is larger than Na atom
13	Which element has lowest melting point	A. Beryllium B. Magnesium C. Calcium D. Barium
14	Which of the following element has lowest ionization energy	A. Beryllium B. Boron C. Carbon D. Oxygen
15	Which of the following statement is correct	A. Na atom is smaller than Na^+ B. Na atom is larger than K atom C. F atom is smaller than F^- D. F atom is larger than F^-

D. F atom is larger than F^{\sup}

16 Which is the longest periodic table

- A. 4
- B. 5
- C. 6
- D. 7

17 Number of elements in the first period of the periodic table is

- A. 2
- B. 8
- C. 14
- D. 18