Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Which one is chlorous acid	A. HClO B. HClO₂ C. HClO₃ D. HClO₄
2	Which is the strongest acid	A. HClO B. HClO₂ C. HClO₃ D. HClO₄
3	The anhydride of $HClO_4$is	A. ClO₃ B. ClO₂ C. Cl₂O₅ D. Cl₂O₇
4	Which halogen will react spontaneously with Au(s) to produce Au^{3+}	A. Br₂ B. F₂ C. I₂ D. Cl₂
5	Hydrogen bond is the strongest between the molecules of	A. HF B. HCl C. HBr D. HI
6	Which one is perchloric acid	A. HClO B. HClO C. HClO₃ D. HClO₄
7	Which is the second most abundant element in the universe	A. H B. He C. CO D. C
8	Which one of halogens is a liquid	A. F₂ B. Cl₂ C. Br₂ D. I₂
9	Out of all the elements of groups VI-A the highest melting and boiling points is shown by the element	A. Te B. Se C. S D. Po
10	The brown gas formed when metal reduces HNO_3	A. N₂O₅ B. N₂O₃ C. NO₂ D. NO
11	Oxidation of NO in air produces	A. N₂O B. N₂O₃ C. N₂O₄ D. N₂O₅
12	In group V-A elements the most electronegative elements is	A. Sb B. N C. P D. As
13	Out of all the elements of Group V-A the highest ionization energy is possessed by	A. N B. P C. Sb D. Bi
14	Which catalyst is used in contact process	A. Fe₂O₃ B. V₂O₅ C. SO₃ D. Ag₂O
15	Maximum number of unpaired electrons is in	A. O₂ B. O₂⁺ C. O₂⁻ D. O₂²⁻

16	What is %age of calcium phosphate in bone ash	A. 20 B. 40 C. 80 D. 60
17	Gold dissolves in "Aqua Regia" due to formation of Halide. Point out correct halide	A. AuF₃ B. AuCl₃ C. AuBr₃ D. AuI₃