

Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Across a period from left to right in the periodic table, the melting and boiling point.	A. Decrease B. Increase C. Remain constant D. First increase upto the middle of period and then decrease
2	The first ionization energy of Na, Mg, Al and Si are in the order of.	A. Na < Mg < Al < Si B. Na > Mg > Al > Si C. Na > Mg < Al < Si D. Na < Mg > Al < Si
3	Which one of the following is a metalloid.	A. sulphur B. Antimony C. Mercury D. Zinc
4	Which one of the following elements has the largest second ionization energy.	A. O B. Na C. F D. Ne
5	Which class of elements shows low value of first ionization potential.	A. Alkali metals B. Alkaline earth metals C. Halogens D. Noble gases
6	As you proceed across a period in the periodic table the first ionization energy	A. Decrease B. Increase C. Remains constant D. First increase up the middle of period and then decreases
7	In a period, from left to right in the periodic table, the size of atom generally.	A. Increases B. decreases C. Remains constant D. First increase upto the middle of period and then decreases
8	In which one of the following sets do all three particles have same number of total electrons	A. F ⁻ , Cl ⁻ , Br ⁻ B. Li ⁺ , Na ⁺ , K ⁺ C. N ³⁻ , O ²⁻ , F ⁻ D. Na ⁺ , Mg ²⁺ , K ⁺
9	In which block of periodic table non metals are present.	A. s B. p C. d D. f
10	Transition elements are those	A. Which comes before uranium B. Which comes after uranium C. Which are prepared artificially D. Which are in between s-block and p - block elements
11	Majority of the elements of the periodic table are.	A. Semi metals B. Non metals C. Metals D. Noble metals
12	What criteria did Mendeleev use in arranging his periodic table.	A. Atomic mass B. Atomic number C. Mass number D. Density
13	What are the total numbers of periods in the modern periodic table.	A. 3 B. 5 C. 7 D. 8
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15	Fluorine is in group VIIA of periodic table. Its chemistry will most closely resembles that of.	A. Argon B. Boron C. Iodine D. Sulphur
16	Fluorine is in group VII A of periodic table. Its chemistry will most closely resembles that of.	A. Argon B. Boron C. Iodine D. Sulphur
17	Which one of the following sets has coinage metaals is it.	A. Cu, Hg, Au B. Cu, Ag, Au C. Ag, Au, Hg D. Cu, Fe, Au