

## Chemistry Fsc Part 2 Online Test

Sr	Questions	Answers Choice
1	Which one of the following polymers is called a polyamide	A. Nylon B. Rayon C. Terylene D. Orlon
2	A polymeric substance that is formed in the liquid state and then hardened to a right solid is called a	A. Fiber B. Plastic C. Varnish D. Polyamid resin
3	Plastics are a pollution problem because many plastics	A. Are made from petroleum B. Are very inflammable C. Burn to produce toxic fumes D. Decompose to produce toxic products
4	In which these process are small organic molecules made into macromolecules.	A. The cracking of petroleum fractions B. The fractional distillation of crude oil C. The polymerization of ethene D. The hydrolysis of proteins
5	Which property of triglycerides is used to determine its molecular mass.	A. acid number B. Saponification number C. Iodine number D. gold number
6	Which process is used to convert vegetable oil to vegetable ghee.	A. Hydrolysis B. Oxidation C. Esterification D. Hydrogenation
7	Fats are	A. Saturated glycerides B. Unsaturated glycerides C. Polyhydroxy ketose D. Polyhydroxy aldose
8	Which amino acid is present in cheese	A. Glycine B. Alanine C. Tyrosine D. Valine
9	Essential amino acids are those which	A. Are present in every protein B. Must be supplied to body through diet C. Contain two carbocyclic acid and one amino group D. Is synthesised by our body
10	Which is simplest amino acid	A. Alanine B. Protein C. Lysine D. Glycine
11	The number of peptide bonds in dipeptide is	A. 0 B. 1 C. 2 D. 3
12	In which process, an amino acid is produced	A. Wurtz synthesis B. Strecker synthesis C. Kolbe synthesis D. Cannizzaro reaction
13	Which one of the following is not amino acid.	A. Alanine B. Glycine C. Aspartic acid D. Aniline
14	Flavour of ethyl butyrate is	A. Orange B. Pine apple C. Banana D. Apricot

15	What is Glacial acetic acid	<p>A. Pure acetic acid B. 95% acetic acid C. a mixture of acetic acid and glycerol D. Vinegar</p>
16	Which substance is used to coagulate rubber latex	<p>A. Ethyl alcohol B. Acetaldehyde C. Acetic acid D. Water</p>
17	Acetic acid forms a dimer in liquid phase because.	<p>A. Low ionization constant of acid B. High solubility in water C. Hydrogen bonding D. Greater polarity</p>