

Chemistry 9th Class English Medium Online Test

Sr	Questions	Answers Choice
1	The atomic radii of the elements in periodic table.	A. Increase from left to right in a period B. Do not change from left to right in a period C. Increase from top to bottom in a group D. Decrease from top to bottom in a group
2	All the elements of Group II A are less reactive than alkali metals. This is because these elements have.	A. Decreased nuclear charge B. Similar electronic configuration C. High ionization energies D. Relatively greater atomic size.
3	Period number of $^{27}_{13}\text{Al}$ is	A. 1 B. 2 C. 3 D. 4
4	Main group elements are arranged ingroups.	A. 7 B. 6 C. 8 D. 10
5	Which of the following element belongs to VIII A.	A. Xe B. Mg C. Br D. Na
6	Which of the following groups contain alkaline earth metals.	A. I A B. II A C. VII A D. VIII A
7	Number of periods in the periodic table are.	A. 7 B. 8 C. 5 D. 16
8	When we move from left to right in a period, atomic size.	A. Increases B. Decreases C. First increases then decreases D. None of the above
9	The element having less value of ionization energy and less value of electron affinity is likely to belong to.	A. Group 1 B. Group 13 C. Group 16 D. Group 17
10	Which element has the highest melting point.	A. K B. Cs C. Na D. Rb
11	Which element is the most reactive element?	A. Fluorine B. Oxygen C. Chlorine D. Nitrogen
12	How many electrons can nitrogen accept in its outermost shell.	A. 2 B. 3 C. 4 D. 5