

Chemistry 9th Class English Medium Online Test

Sr	Questions	Answers Choice
1	A bond pair is covalent molecules usually has.	A. One electron B. Two electron C. Three electron D. Four electron
2	A bond formed between two non metals is expected to be	A. Ionic B. Coordinate covalent C. Metallic D. Covalent
3	Identify the covalent compound	A. NaCl B. H_2O C. KF D. MgO
4	In the formation of AlF_3 , aluminum atom loses.....electrons.	A. 1 B. 4 C. 3 D. 2
5	Silicon belongs to Group IVA . It haselectrons in the valence shell	A. 2 B. 6 C. 3 D. 4
6	How many covalent bonds do N_2 molecule have	A. 3 B. 4 C. 2 D. 5
7	Number of electrons in nitrogen molecule is.	A. 2 B. 4 C. 6 D. 8
8	Which types of attractive forces are present in ionic compounds.	A. Covalent bonds B. Electrostatic forces of attraction C. Metallic bonds D. Coordinate covalent bonds
9	During the formation of ionic bond heat is.	A. Remains same B. Absorbed C. Released D. Both a and b
10	How many valence shell electrons are there in Na^+ ion.	A. 8 B. 9 C. 1 D. 10
11	The electropositive elements have the tendency to	A. Lose electrons B. Gain electrons C. Share electrons D. All of these
12	How many electrons are there in the valence shell of sodium atom.	A. One B. Two C. Three D. Four