

Chemistry 9th Class English Medium Online Test

Sr	Questions	Answers Choice
1	The boiling point of alcohol is	A. 44°C B. 78°C C. 53°C D. 19°C
2	The boiling point of water is	A. 0°C B. 100°C C. 35°C D. 25°C
3	The force of attraction between water molecules is.	A. Ionic bonding B. Covalent bonding C. Hydrogen Bonding D. Coordinate Covalent bonding
4	Which of the following is an example of polar covalent compound.	A. Cl_2 B. H_2 C. O_2 D. HCl
5	Hydrogen bonding is always found in	A. Non-polar molecules B. Homonuclear molecules C. Polar Molecules D. All of the above
6	Metals have the tendency to lose electrons due to.	A. High ionization energies B. Low ionization energies C. Low electron affinity D. None of the above
7	Malleability is the property by virtue of which a metal can be drawn into.	A. Rods B. Plates C. Sheets D. Wires
8	In metals, the hold of nucleus over the valence shell electrons is weak due to.	A. High electron affinity B. Large sized atoms C. High ionization energies D. All of the above
9	Which types of bond is present between NH_3 and BF_3	A. Covalent Bond B. Ionic Bond C. Coordinate covalent bond D. Metallic Bond
10	How many lone pairs are present on nitrogen in ammonia molecule.	A. One B. Two C. Three D. Four
11	Which one of the following is an electron deficient molecule.	A. NH_3 B. O_2 C. BF_3 D. N_2
12	Dative covalent bonds are also known as	A. Covalent bond B. Ionic Bond C. Metallic Bond D. Coordinate covalent bond