

## Chemistry 9th Class English Medium Online Test

1 The colour of anhydrous copper (Il)sulphate solid is 2. Black C. White D. Blue C. Forward C. Moderate D. Perduct C. Moderate D. Negligible C. All the reactants cover into products C. All the reactants cover into products C. All the reactants cover into products C. All the reactants cover into D. None of these C. All the reactants cover into D. None of these C. All the reactants cover into D. None of these C. Products D. None of these C. Products D. None of these C. Products D. Reactants C. Products D. None reaction C. Reversible Reaction C. Reversible Reaction D. Addition reactions D. Addition reactions D. Addition reactions D. Addition reactions D. Proceeds from left to ig. C. In which the products do not reombine to form reatants are called C. In reversible reactions D. Proceeds from left to ig. C. In which reactants reactions D. Slow down gradually D. Proceeds from left to ig. C. In which reactants reactions D. Slow down gradually D. Proceeds from left to ig. C. In which reactants reactions D. Slow down gradually C. Productive reactions D. Sl	
2 The new substance formed in a chemcial reaction is.  2 Forward D. Product 3 In the beginning the rate of reverse reaction is.  3 In the beginning the rate of reverse reaction is.  4 A Slow B. Very fast C. Moderate D. More of these D. All the reactants covert int D. B. Left to right D. None of these D. None reaction D. Reversible D. None reaction D. Reversible reactions D. Irreversible reactions D. Irreversible reactions D. Irreversible reactions D. A Reversible reactions D. Reversible reac	
3	
4 A complete reaction is in which  B. All the reactants covert in products C. All the reactants do not opposite to products D. Half reactants covert into products D. Half reactants covert into the products of the substances that combine are called.  A. Right to left B. Left to right C. Both a and b. D. None of these A. Masses B. Materials C. Products D. Reactants  A. Dynamic B. Irreversible C. Reversible C. Reversible D. Non- reactive  A. Decomposition reaction B. Forward reaction C. Reverse reactions D. Irreversible reactions D. Irreversible reactions D. Irreversible reactions D. Addition reactions D. Addition reactions D. Addition reactions D. reversible reactions D. slow down gradually B. Proceeds from left to rig C. in which reactants react products D. Slow down gradually A. Product never recombine reactants	
The forward reaction takes place from  C. Both a and b D. None of these  A. Masses B. Materials C. Products D. Reactants  A. Dynamic B. Irreversible C. Products D. Reactants  A. Dynamic B. Irreversible C. Reversible D. Non- reactive  A. Decomposition reaction B. Forward reaction C. Reversible D. Non- reactive  A. Decomposition reaction B. Forward reaction C. Reversible reactions D. Irreversible reactions D. Addition reactions D. Addition reactions D. Addition reactions D. reversible reactions D. Sow down gradually A. Product nver recombine reactants P. Slow down gradually A. Product nver recombine reactants P. Slow down gradually A. Product nver recombine reactants P. Slow down gradually A. Product nver recombine reactants	nto covrt into
6 In chemical reaction, the substances that combine are called.  7 Such reaction which continue in both directins are called.  8 Which type of reactions speed up gradully?  8 Which type of reactions speed up gradully?  9 The reaction in which the products can recombine to formreactants are called.  10 The reaction in which the products do not reombine to form reatants are called.  11 A reverse reaction is one that  12 A reverse reaction is one that  13 A reverse reaction is one that  14 A reverse reaction is one that  15 B. Materials C. Products D. Reactants are called.  16 A Decomposition reaction in B. Forward reaction B. Forward reaction C. Reverse reactions D. Addition reactions D. Addition reactions B. Decomposition reactions D. reversible reactions D. reversible reactions D. reversible reactions D. reversible reactions D. Reverse reaction left to rig C. In which reactants react products D. Slow down gradually A. Product nver recombine reactants	
Such reaction which continue in both directins are called.  B. Irreversible C. Reversible D. Non- reactive  A. Decomposition reaction B. Forward reaction C. Reverse reactions D. Irreversibel reactions  Proverse reactions D. Irreversible reactions  The reaction in which the products can recombine to formreactants are called.  The reaction in which the products do not reombine to form reatants are called.  A. Reversible Reaction B. Irreversible reactions C. Decomposition reactions D. Addition reactions D. Addition reactions B. Decomposition reactions C. Irreversible reactions D. reversible reactions D. Speeds up gradually B. Proceeds from left to rig C. In which reactants react products D. Slow down gradually A. Product nver recombine reactants	
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B. Proceeds from left to rig C. In which reactants react products D. Slow down gradually  A. Product nver recombine reactants	าร
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The characteristics of reversible reactions are the following except.  B. They never complete C. They have a double arro between reactants and pro D. The proceed in both way	row oducts