

## Chemistry 10th Class English Medium Online Test

Sr	Questions	Answers Choice
1	A substance which can behave as an acid as well as a base is called.	A. Acid B. Base C. Amphoteric D. Neutral
2	A conjugate base is a specie formed by donating a.	A. Proton B. Electron pair C. Neutron D. Electron
3	According to Bronsted and Lowry concept a base is a substance that can accept.	A. Proton B. Electron pair C. Neutron D. Electron
4	A conjugate acid is a specie formed by accepting a.	A. Proton B. Electron pair C. Neutron D. Electron
5	According to Bronsted and Lowry concept an acid is a substance that can donate.	A. Proton B. Electron pair C. Neutron D. Electron
6	Bronsted and Lowry presented their theories of acids and bases in	A. 1785 B. 1787 C. 1923 D. 1925
7	Which one is not an Arrhenius base?	A. NaOH B. KOH C. $\text{Ca(OH)}_2$ D. $\text{NH}_3$
8	Which one is not an Arrhenius acid?	A. HCl B. $\text{H}_2\text{SO}_4$ C. $\text{CO}_2$ D. $\text{HNO}_3$
9	According to Arrhenius concept acid is a substance which dissociates in aqueous solution to give.	A. Hydrogen ions B. Hydroxide ions C. Both a and b D. None of these
10	Arrhenius presented his concept about acids and bases in.	A. 1785 B. 1787 C. 1923 D. 1930
11	All bases turn red litmus	A. Red B. Blue C. Pink D. White
12	All acids turn blue litmus.	A. Red B. Blue C. Pink D. White