

## Chemistry 10th Class English Medium Online Test

Sir         Questions         Answers Choice           1         Formic acid is present in         A String of Dies B. Sour milk C. Appile D. Pats           2         Citric acid is present in         A Citric acid is present in           3         Which acid used for food preservation?         A Houb 2 / sub > SO / sub > 4 / sub > D. Appile           4         Which acid is used for etching designs on copper plates?         A Hosub > 2 / sub > SO / sub > 4 / sub > D. CH2 in the sub of sub > 4 / sub > 4			
1 Formic acid is present in B. Sour milk C. Apple D. Fats  2 Citric acid is present in Citric faults B. Sour milk C. Rapple D. Fats C. Apple D. Fats C. Apple D. Fats C. Rapple D. R	Sr	Questions	Answers Choice
2 Citric acid is present in  8. Sour milk C. Rancia butter D. Apple  3 Which acid used for food preservation?  4 Which acid used for etching designs on copper plates?  4 Which acid is used for etching designs on copper plates?  5 Which acid is used an electrolyte in lead storage battery?  6 Which acid is used an electrolyte in lead storage battery?  6 Which one of the mineral acid.  7 When acid reacts with sulphites and Bi sulphates which gas is evolved?  8 When acid react with carbonates and bicarbonates which gas is evolved?  9 When acid react with metals which gas is evolved?  10 Which one is Lewis Acid?  11 A product of any Lewis acid base reaction is a single specie called.  A Proton B. Besub-2-4 sub-2-4	1	Formic acid is present in	B. Sour milk C. Apple
8. HNO-sub>3 (sub> C. HOI D. CH-sub>3 (sub> C. HOI D. CH-sub>3 (sub> COOH D. CH-sub>3 (sub) (sub	2	Citric acid is present in	B. Sour milk C. Rancid butter
4 Which acid is used for etching designs on copper plates?  5 Which acid is used an electrolyte in lead storage battery?  A H <sub>2</sub> SO <sub>4</sub> COH C HCI D. CH <sub>3</sub> COH B HNO/sub>3COH B HSub>2SO <sub>4</sub> COH C HCI D. CH <sub>3</sub> C HCI D. CH <sub>3</sub> COH C HCI D. CH <sub>3</sub> C HCI D. CH <sub>3</sub> COH C HCI D. CH <sub>3</sub> C HCI D. CH <sub>3</sub> C HCI D. CH <sub>3</sub> C HCI D. CH <sub>2</sub> C HCI D. CH <sub>3</sub> C HCI D. CH <sub>2</sub> C HCI D. CH C HC	3	Which acid used for food preservation?	B. HNO <sub>3</sub> C. HCl
5 Which acid is used an electrolyte in lead storage battery?  B. HNCssub-33-/sub>CC CHCl D. CH-sub>3-/sub>COOH  A. HCl B. Hssub>2-/sub>SO-sub>4-/sub> C. HNOssub-33-/sub> D. All of these  7 When acid reacts with sulphites and Bi sulphates which gas is evolved?  B. CO-sub-2-/sub- C. SO-sub-2-/sub- D. NH-sub-3-/sub- D. NH-sub-3-/sub- D. NH-sub-3-/sub- D. N-sub-2-/sub- D. N-sub-2-/	4	Which acid is used for etching designs on copper plates?	B. HNO <sub>3</sub> C. HCI
B. H-sub->28 (NO <sub->2</sub-> 10 (NO <sub->3</sub-> 10 (NO <su< td=""><td>5</td><td>Which acid is used an electrolyte in lead storage battery?</td><td>B. HNO<sub>3</sub> C. HCl</td></su<>	5	Which acid is used an electrolyte in lead storage battery?	B. HNO <sub>3</sub> C. HCl
When acid reacts with sulphites and Bi sulphates which gas is evolved?  8 When acid react with carbonates and bicarbonates which gas is evolved?  8 CO <sub>2</sub> C. SO <sub>2</sub> D. NH <sub>3</sub> C. Cl <sub>2</sub> C. Cl <sub>2</sub> C. Cl <sub>2</sub> D. NSub>2 D. NSub>2 D. NSub>2 C. Cl <sub>2</sub> D. NSub>2 D. NSub>2 C. Cl <sub>2</sub> C. Cl <sub>2</sub> D. NSub>2 C. Cl <sub>2</sub> D. NSub>3 D. NSub>3 D. All of these  11 A product of any Lewis acid base reaction is a single specie called.  A Proton B. Electron pair C. Neutron  A Proton B. Electron pair C. Neutron	6	Which one of the mineral acid.	B. H <sub>2</sub> SO <sub>4</sub> C. HNO <sub>3</sub>
8 When acid react with carbonates and bicarbonates which gas is evolved?  9 When acids react with metals which gas is evolved?  A. H <sub>2</sub> B. C <sub>2</sub> A. H <sub>2</sub> C. Cl*sub>2 B. O <sub>2</sub> C. Cl*sub>2 C. Cl*sub>2 C. Cl*sub>2 D. N\sub>2 C. Cl*sub>2 D. N\sub>3 D. N\sub>3 C. FeCl*sub>3 C. FeCl*sub>3 D. All of these  11 A product of any Lewis acid base reaction is a single specie called.  A. Salt B. Water C. Adduct D. None of these  A. Proton B. Electron pair C. Neutron	7	When acid reacts with sulphites and Bi sulphates which gas is evolved?	B. CO <sub>2</sub> C. SO <sub>2</sub>
9 When acids react with metals which gas is evolved?  10 Which one is Lewis Acid?  A. BF <sub>2</sub> D. N <sub>2</sub> A. BF <sub>3</sub> B. AlCl <sub>3</sub> C. FeCl <sub>3</sub> D. All of these  A. Salt B. Water C. Adduct D. None of these  A. Proton B. Electron pair C. Neutron	8	When acid react with carbonates and bicarbonates which gas is evolved?	B. CO <sub>2</sub> C. Cl <sub>2</sub>
10 Which one is Lewis Acid?  B. AlCl <sub>3</sub> C. FeCl <sub>3</sub> D. All of these  A. Salt B. Water C. Adduct D. None of these  A. Proton B. Electron pair C. Neutron	9	When acids react with metals which gas is evolved?	B. O <sub>2</sub> C. Cl <sub>2</sub>
A product of any Lewis acid base reaction is a single specie called.  B. Water C. Adduct D. None of these  A. Proton B. Electron pair C. Neutron	10	Which one is Lewis Acid?	B. AlCl <sub>3</sub> C. FeCl <sub>3</sub>
12 According to Lewis concept a base is a substance which can donate.  B. Electron pair C. Neutron	11	A product of any Lewis acid base reaction is a single specie called.	B. Water C. Adduct
	12	According to Lewis concept a base is a substance which can donate.	B. Electron pair C. Neutron