

ECAT Physics Chapter 20 Atomic Spectra Online Test

Sr	Questions	Answers Choice
1	If the distance between two charges is doubled, the force between them will become:	A. Double B. Half C. One third D. One fourth
2	The inkjet printer ejects a thin stream of:	A. Water B. Oil C. Ink D. Any above E. None of these
3	Selenium is:	A. An insulator B. A conductor C. Both A and B D. Excellent conductor E. None of these
4	The photocopying process is called:	A. Geography B. Sonography C. Xerography D. Zerography E. None of these
5	The electric field lines start from:	A. Positive charge B. Negative charge C. Either A and B D. Neutron E. An atom
6	Static electricity is produced by the transfer of:	A. Electrons B. Protons C. One fluid D. Two fluid E. None of these
7	Coulomb multiplied by volt by volt gives the unit called:	A. farad B. Ohm C. Second D. joule E. Watt
8	Gaussian surface is always:	A. Rectangular B. Spherical C. Cylinder D. Box shape E. Any of these
9	Photons must have energy equal to	A. ev B. En C. hf D. None of these
10	Graph of Black body radiation is example of	A. Band spectra B. Continuous spectra C. Line spectra D. All
11	Ultraviolet region lies in _____ series	A. Lyman B. Balmer C. Paschen D. Brackett
12	We can excite an atom by	A. Bombardment of particles B. Radiating photons C. Providing potential difference D. All answers are true
13	The life time of metastable state is equal to	A. Life time of excited state B. Greater than by excited state C. Zero D. Less than by excited state
14	The first shell near the nucleus is	A. L-shell B. K-shell C. N-shell D. M-shell

		D. M-shell
15	CT scanning is the abbreviated name of	A. Computed Technology B. Computed Technique C. Computed Technology D. Computerized Technique
16	The minimum wavelength of X-rays produced of 1KV potential difference is applied across the anode and cathode of the tube is	A. 1.24×10^{-10} m B. 7.92×10^{-20} m C. 2.78×10^{-14} m D. 3.88×10^{-11} m
17	X-ray are also known as	A. Roentgen rays B. Maxwell rays C. Plank range D. Einstein rays
18	In flesh, light element like carbon, hydrogen and oxygen predominate. Three elements allows _____ amount of incident X-ray to pass through them	A. Small B. Greater C. Equal D. Sometimes
19	The value of the metastable state for Neon is	A. 20.66eV B. 20.61eV C. 19.23eV D. 18.70eV
20	In helium Neon Laser Neon = 15% and Helium = 85% used. The lasing gas this unit is	A. Helium B. Neon C. Both D. None of these
21	X-rays produced in a tube operating at 10^5 V. The speed of X-rays produced is	A. 3×10^8 m/s B. 3.1×10^8 m/s C. 2.8×10^8 m/s D. 1.88×10^8 m/s
22	Consider a photon of continuous X-ray and a photon of characteristics X-ray of same wavelength. Which of the following is/are different for the two photons	A. Frequency B. Penetrating power C. Energy D. Method of creation
23	Atoms of hydrogen gas can be excited by passing electric current through it when the gas is filled into the discharge tube at a pressure which is	A. Less than atmospheric pressure B. Much less than atmospheric pressure C. Greater than atmospheric pressure D. Much greater than atmospheric pressure E. Both C and D
24	The range of wavelengths of colours in the visible colours is	A. 140 nm to 456 nm B. 10 nm to 56 nm C. 410 nm to 656 nm D. 910 nm to 956 nm E. None of these
25	The results of spectra obtained by Blamer were expressed in 1896 by	A. Bohr B. Rydberg C. Planck D. Rutherford E. Coulomb
26	The first series which was identified in the spectrum of hydrogen is called:	A. Lyman series B. Balmer series C. Paschen series D. Brackett series E. Pfund series
27	Balmer series was identified in:	A. 1685 B. 1785 C. 1885 D. 1985 E. 1585
28	Balmer series lies in that region of electromagnetic wave spectrum which is called:	A. Visible region B. Invisible region C. Infra-red region D. ultraviolet region E. None of these
29	The natural arrangement of colours in the spectrum of white light spectrum is	A. VIBGYOR B. ROYBGIV C. ROYBIGV D. BIGROYV E. None of these
		A. Bohr B. Rutherford

30	The results of spectra obtained by Balmer were expressed in 1896 by:	<p>B. Rydberg</p> <p>C. Planck</p> <p>D. Rutherford</p> <p>E. Coulomb</p>
31	The process of formation of spectrum is called:	<p>A. Interference</p> <p>B. Spectroscopy</p> <p>C. Dispersion</p> <p>D. Reflection</p> <p>E. Botha (A) and (D)</p>
32	Spectrum represents the number of component colours present in certain light in terms of:	<p>A. Wavelength</p> <p>B. Frequency</p> <p>C. Energy</p> <p>D. Both (A) and (B)</p> <p>E. All of these</p>
33	Tick the series which lies in the visible region:	<p>A. Lyman series</p> <p>B. Balmer series</p> <p>C. Paschen series</p> <p>D. Brackett series</p> <p>E. P fund series</p>
34	Tick the series which lie/s in. the infra-red region.	<p>A. Pfund series</p> <p>B. Brackett series</p> <p>C. Paschen series</p> <p>D. All of these</p> <p>E. None of these</p>
35	Lyman series in the spectrum of hydrogen exists in the :	<p>A. Infra-red region</p> <p>B. Visible region</p> <p>C. Ultraviolet region</p> <p>D. Both(A) and (B)</p> <p>E. None of these</p>
36	The formula of Brackett series can be obtained by putting in the general formula, the value of n equal to:	<p>A. one</p> <p>B. two</p> <p>C. three</p> <p>D. four</p> <p>E. five</p>
37	An electron of the hydrogen atom in the second orbit is called its:	<p>A. Ground state</p> <p>B. Excited state</p> <p>C. Ionized state</p> <p>D. Any of these</p> <p>E. None of these</p>
38	Energy required by an electron revolving in certain orbit to jump to an excited state is called:	<p>A. Ionization energy</p> <p>B. Ionization potential</p> <p>C. Excitation energy</p> <p>D. Excitation potential</p> <p>E. None of these</p>
39	The transitions of electrons in the hydrogen atom result in the emission of spectral lies in the:	<p>A. Ultra red region</p> <p>B. Visible region</p> <p>C. Ultraviolet region</p> <p>D. Any of these</p> <p>E. None of these</p>
40	The holes created in the L and M shells are occupied by transitions of:	<p>A. Electrons from lower states</p> <p>B. Electrons from higher state</p> <p>C. Positrons from higher states</p> <p>D. Electrons from K shell</p> <p>E. Both (A) and (B)</p>
41	Braking radiation causes:	<p>A. Continuous spectrum</p> <p>B. Line Spectrum</p> <p>C. Band spectrum</p> <p>D. Discrete spectrum</p> <p>E. All of these</p>
42	In case of braking radiations, when the rate of deceleration is very large, the emitted radiation corresponds to:	<p>A. Short wavelength</p> <p>B. Large wavelength</p> <p>C. Very large wavelength</p> <p>D. Low frequency</p> <p>E. Both (B) and (C)</p>
43	X-rays can penetrate in a solid matte through a distance of several:	<p>A. Kilo metres</p> <p>B. Metres</p> <p>C. Centimeters</p> <p>D. A few angstroms</p> <p>E. One micrometer</p>
44	An compared to solid matter, a crack or an air bubble allows:	<p>A. Great amount of X-rays to pass</p> <p>B. Smallest amount of X-rays to pass</p> <p>C. Very small amount of X-rays to pass</p> <p>D. Any of these</p> <p>E. None of these</p>

45	The shell closer to the nucleus is called:	A. N shell B. L shell C. K shell D. M shell E. O shell
46	By CAT scans, we can detect the density difference of the order of:	A. 1% B. 20% C. 30% D. 50% E. 70%
47	Laser is a beam of:	A. Visible light B. Infra red light C. Ultra violet light D. Violet light only E. yellow light only
48	A metastable state:	A. Is an excited state B. Is that in which excited electron is stable C. Is that in which excited electron is usually unstable D. Means a time interval of 10^{-8} second E. Both (A) and (C)
49	The He-Ne laser discharge tube is filled with:	A. 85% He B. 15% He C. 50% He D. 60% He E. 85% Ne
50	The spectrum emitted from hydrogen filled discharge tube is:	A. Line spectrum B. Discrete spectrum C. And spectrum D. Absorption spectrum E. Both (A) and (B)
51	The lasing or active medium in He-Ne laser discharge tube is:	A. Nitrogen B. Helium C. Hydrogen D. Neon E. None of these