

General Science 8th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Carbon burns in air to release energy along with the formation of.	A. Carbon dioxide and hydrogen B. Carbon monoxide and hydrogen C. Carbon di oxide D. Carbon dioixe and water
2	The product fo the reaction between zinc and dilute sulphuric acid are.	A. Zinc oxide and water B. Zinc sulphide and water C. Zinc sulphate and hydrogen D. Zinc sulphide and hydrogen
3	Heating of solid potassium chlorate produces a gas.	A. Chlorine B. Carbon dioxide C. Carbon mono oxide D. Oxygen
4	How many oxygen atoms are present in onw molecule of $Mg(HCO_3)_2$	A. 2 B. 4 C. 6 D. 8
5	The reaction between calcium oxide and carbon dioxide to form calcium carbonate is an example of.	A. Decomposition reaction B. Addition reaction C. Acid-Base reaction D. Neutralizaion reaction
6	Thermal decompositionof calcium carbonate produces a gas.	A. Oxygen B. Carbon di oxide C. Nitrogen D. Carbon monoxide
7	Substances whihc take part in a chemical reaction are called.	A. Products B. Ractants C. Elements D. Compunds
8	When methane is burnt in air, CO_2 and..... are formed.	A. O_2 B. H_2O C. Glucose D. Energy
9	During photosynthesis in plants, carbon dioxide CO_2 and water react to produce.	A. Water B. Energy C. Glucoe D. CO_2
10	Symbol and formulae of the reactants are written on..... side fo arrow.	A. Left B. Right C. Upward D. Both side
11	The equation in which the number of atoms of each element on both sides are not equal is called.	A. Quadratic equation B. Unbalanced equation C. Balanced equation D. Standard equation
12	Law fo conservation of mass was put forward by	A. Newton B. Plank's C. Lavoisier D. Einstein
13	Law of conservation of mass put forward in.	A. 1783 B. 1784 C. 1785 D. 1786
14	Fire works is an example of reactions.	A. Addition B. Exothermic C. Endothermic D. Decomposition
15	During a chemical reaction, total mass of the product is equal to the total mass of reactants, this is a statements of.	A. Law of conservation of momentum B. Law of conservation of energy C. Law of conseration of mass D. Law of conseration of acceleration

16	The chemical reaction during which a compound splits up into two or more simple substances, is called.	A. Addition reaction B. Decomposition C. Replacement reaction D. Balanced reaction
17	Exothermic reaction are those reactions during which heat is.	A. Given out B. Remain same C. Absorbed D. Used