

General Science 7th Class English Medium Chapter 4 Online Test

Sr	Questions	Answers Choice
1	The maximum numebr of electrons that N-Shell accomodate are.	A. 2 B. 8 C. 18 D. 32
2	Element present in grops have same number of in their outermost shell.	A. Electron B. Proton C. Atoms D. Neutron
3	An element with fifteen protons in its atomic nucleus has atomic numebr.	A. 5 B. 10 C. 15 D. 20
4	There are groups in periodic table.	A. 2 B. 8 C. 18 D. 32
5	Positively charged particle of an atom is called.	A. Electron B. Proton C. Atom D. Neutron
6	The mass of an atom is almost due to the.	A. Electron B. Proton C. Nucleus D. Neutrons
7	The central part of an atom is called.	A. Electron B. Proton C. Neutron D. Nucleus
8	The negetively charged particle arod nucleus are called.	A. Electron B. Proton C. Atom D. Neutron
9	Area around nucleus where electrons are found are called.	A. Orbit B. Shell C. Both a and b D. None of these
10	The total numebr of electrons which M shell can accommodate	A. 2 B. 8 C. 18 D. 32
11	The numebr of electorns present in an atom s equal to number of.	A. Nucleus B. Neutrons C. Proton D. Atom
12	Atomic number of element presetnin period 2 and group 15 of the periodic table.	A. 4 B. 5 C. 6 D. 7
13	The particle of an atom having negative charge on it.	A. Electron B. Proton C. Neutron D. Nucleus
14	Letter K is used to express theshell around the nucleus of an atom.	A. 1st B. 2nd C. 3rd D. 4th
15	The elements of group 18 are called.	A. Alkali metals B. Noble gases C. Alkaline metals D. Transition metals

16	Formula showing maximum number of electrons in a shell.	A. n2 B. 2n2 C. 3n2 D. 4n2
17	In elements are arranged in order of increasing atomic numbers.	A. Group B. Period C. Orbit D. Shell
18	The particles of an element having different mass numbers.	A. Proton B. Neutrons C. Isotopes D. Atoms
19	The charge present on atom is	A. Positive B. Negative C. Both a and b D. Neutral
20	How many periods are there is the Periodic table.	A. Five B. Six C. Seven D. Eight
21	are fundamental particles of an atom.	A. Electron B. Proton C. Neutron D. All of these